

## Suspended particle effects on ClO<sub>2</sub>/ultraviolet light combined disinfection of effluent

WANG Jian-ling<sup>1,\*</sup>, WANG Bao-zhen<sup>1</sup>, WANG Lin<sup>2</sup>, ZHANG Jin-song<sup>3</sup>, HUANG Wen-zhang<sup>3</sup>

(1. School of Municipal and Environmental Engineering, Harbin Institute of Technology, Harbin 150090, China. E-mail: wangjianling@sina.com.cn; 2. School of Environmental Science and Engineering, Ocean University of China, Qingdao 266003, China; 3. Shenzhen Water (Group) Co., LTD., Shenzhen 518031, China)

**Abstract:** The concentration of suspended solids of effluent often varies in a wide range, therefore the dose of ultraviolet light (UV) in disinfection process needs to be adjusted to meet the disinfection criterion at a high frequency, and the desired disinfection effect is difficult to be ensured. The particles size and particle-associated fecal coliform (F.C.) contribution, and their influence on UV disinfection were investigated when ClO<sub>2</sub> and UV combined disinfection process was used. The results showed that suspended solids content had a major impact on UV disinfection efficiency, especially the large particle size fraction. Particles ( $D > 10 \mu\text{m}$ ) associated F. C. were difficult to be disinfected and were the main part of the tailings of F.C. inactivation curve. Pre-ClO<sub>2</sub> oxidation could reduce the number of particles in effluent, and make large particles decrease to small ones. Therefore, the influence of particles on UV disinfection could be reduced after pre-ClO<sub>2</sub> oxidation, and the resistance ability to particle loadings of combined process was enhanced. Moreover, the combined process has a lot of advantages, such as low toxicity, low operational/maintenance costs; it is also convenient to be established in the existing wastewater plant or the new planned one.

**Keywords:** effluent disinfection; bio-toxicity; UV disinfection; ClO<sub>2</sub>/UV combined disinfection

### Introduction

The secondary effluent of wastewater treatment plant should be disinfected to eliminate pathogenic protozoan cysts, bacteria species, and virus, before treated effluent discharge or reuse. UV disinfection has gained considerable popularity due to the fact that it imparts only energy to the water or wastewater without producing quantifiable amounts of known disinfectant by-products, as do chemical disinfectants (Ronald *et al.*, 2003; White, 1999).

Some studies (Emerick *et al.*, 1999; Frank *et al.*, 2002) show that the suspended solids content of wastewater has a major impact on UV disinfection efficiency. Because the activated sludge floc flows with the influent, and is in suspension in secondary clarifiers, suspended solids concentration in effluent often varies in a wide range. To satisfy disinfection criteria, UV disinfection doses should be adjusted, and as a result UV process is highly operational and maintenance-sensitive. Prior particles removal by coagulation or filtration is often used in wastewater treatment plants in west countries, if disinfected with UV process (Chiu *et al.*, 1999; Andreadakis *et al.*, 1999). In this case, more space and investment are required. Now, the disinfection by UV light radiation has not been widely used in China. Almost all of the constructed wastewater treatment plants in operation have used Cl<sub>2</sub> or ClO<sub>2</sub> disinfection process, and the Cl<sub>2</sub> or ClO<sub>2</sub> facilities have been installed. However, because of operation costs, many wastewater disinfection processes have not be run normally (Cheng *et al.*, 2003). Furthermore, awareness of the adverse health and environmental effects associated

with Cl<sub>2</sub> or ClO<sub>2</sub> has led to serious consideration of doses control, which is negative to ensure the disinfection effect. All the above shows that it is in urgent need to find a high efficiency, relatively inexpensive and safe disinfection process for effluent.

In this paper, the ClO<sub>2</sub>/UV disinfection process was brought forward. Pre-ClO<sub>2</sub> addition was used to oxidize the organic constituents within particles, and transform the large particle size fraction of wastewater to small fraction. Particles counting and size distribution measurements were conducted to investigate the relationship between particle size and UV disinfection effects. Furthermore, particle size contribution, particle-associated fecal coliform (F.C.) contribution of wastewater and disinfection efficiency of ClO<sub>2</sub> disinfection, UV and ClO<sub>2</sub>/UV were studied respectively.

## 1 Materials and methods

### 1.1 Effluent characterization

The test water was taken from the effluent of Shenzhen Binhe WWTP using A/O process (300000 m<sup>3</sup>/d). Table 1 shows the main wastewater characteristics. The experiment was conducted from February to August in 2005.

**Table 1** Characteristics of the effluent from Binhe WWTP using A/O process

Parameter	COD, mg/L	NH <sub>4</sub> -N, mg/L	SS, mg/L	F.C., MPN/L	pH	Water temperature, °C
Aver.	28	18	13	$2.4 \times 10^6$	7.1	26
Min.	20	14	7	$7.2 \times 10^5$	6.8	24
Max.	38	22	18	$5.4 \times 10^6$	7.5	28

Notes: SS. Suspended solid; F.C. fecal coliform; MPN. most probable number

\*Corresponding author

## 1.2 Experimental procedure

A collimated beam apparatus was specially constructed. A diagram of the collimated beam apparatus is shown in Fig.1. The UV dose and depth-averaged UV intensity within the Petri dish sample can be computed based on Beer's law (George and Franklin, 2003).

$$D = I_{ave} \times t \quad (1)$$

$$I_{ave} = I_0 \times \frac{(1 - e^{-2.303 Ad})}{2.303 Ad} \quad (2)$$

where  $D$  is the UV dose, mJ/cm<sup>2</sup>;  $I_{ave}$  is the average UV intensity, mW/cm<sup>2</sup>;  $t$  is the exposure time, s;  $I_0$  is the average incident UV intensity at the water surface of the sample, mW/cm<sup>2</sup> (detected before test every time by UV intensity detection instrument);  $A$  is the absorbance, cm<sup>-1</sup>;  $d$  is the depth of sample, cm.

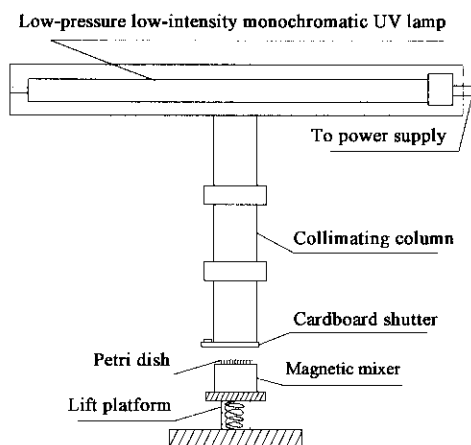


Fig.1 Diagram of collimated beam apparatus

Different doses of ClO<sub>2</sub> (0–15 mg/L) were added to effluent of A/O process, and mixed by magnetic mixer. At the end of contact time (30 min in this study), residual ClO<sub>2</sub> was detected, and a little excessive NaSO<sub>3</sub> was used to end up the disinfection reaction. Then, the samples by ClO<sub>2</sub> disinfected continued to be disinfected by UV apparatus (0–20 mJ/cm<sup>2</sup>). Bio-toxicity and F.C. number of all samples were analyzed. The filtration/re-suspension technique was used to separate the particles into different size classes (Berman *et al.*, 1988). Different size filters (0.45–50 μm) were chosen for sequential filtration. Washing and rinsing solution, used to re-suspend particles, was generated by 0.45 μm filter, and disinfected at high temperature. The particles and their associated bacteria captured on each filter were re-suspended and mixed uniformly by the magnetic mixer.

## 1.3 Materials and analytical procedures

The stock ClO<sub>2</sub> solution was produced by mixing a hydrochloric acid solution and a sodium chlorite solution. The stock solution of ClO<sub>2</sub> was stored in dark at 4°C and was standardized daily. Sodium thiosulfate (Na<sub>2</sub>SO<sub>3</sub>) solution of analytical grade, prepared every

day, was used to dechlorinate ClO<sub>2</sub> residuals. Other chemical solutions were all prepared by analytical grade products and double deionized water.

The toxicity of samples was analyzed by reading light produced by luminescent bacteria (SDI, U.S.) after exposure to a test sample and the result was compared with the light output of a reagent blank. The degree of light loss percentage (an indication of metabolic inhibition in the test organisms) indicated the relative toxicity of the sample, classified into four grades as shown in Fig.2 (John and Ernest, 1999; Kaiser and Palabrica, 1991). Particle counting and size distribution were detected by particle counting apparatus (measuring capacity: 2 μm, IBR, U.S.). Residual ClO<sub>2</sub> was determined by ClO<sub>2</sub> residuals analyzer (measuring capacity: 0.05 mg/L). Faecal coliform (F.C.) number and the chemical oxygen demand (COD) were measured according to the standard methods (U.S. APHA, 1998). All the analyses were started to make within 30 min after sample collection.

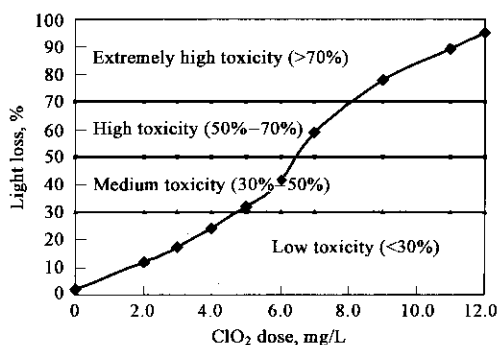


Fig.2 Effect of ClO<sub>2</sub> dosage on bio-toxicity of wastewater  
F.C.=3.2 × 10<sup>6</sup> MPN (most probable number)/L; SS=16 mg/L;  
T= 26.5°C

## 2 Results and discussion

### 2.1 Particles and associated F.C. distribution

Particle counting and size distribution, and F.C. distribution are shown in Table 2, from which it was found that 95% of all the particles have the size smaller than 10 μm. Of all the particles, 3–5 μm particles account for the most, and 2–3 μm particles take the second place. F.C. associated with particles smaller than 10 μm (free F.C. included) was 3.2 × 10<sup>6</sup>, about 98% of all F.C.

### 2.2 Particle size effect on performance of UV disinfection

The influence of different size particles is listed in Fig.3 (UV dose was 20 mJ/cm<sup>2</sup>). F.C. in samples filtered by particle size  $D=0.45$  μm filters was considered to be free ones. The disinfection efficiency was calculated as the negative logarithm of the ratio of F.C. after disinfection ( $N_t$ ) to that before disinfection ( $N_0$ ).

Fig.3 shows that the larger particle size fraction

**Table 2** Distribution of particles and faecal coliform (F.C.) of secondary effluent

Particles size, $\mu\text{m}$	<2 <sup>a</sup>	2—3	3—5	5—10	10—20	20—30	30—50	>50	Total
SS distribution, mg/L	5	3	4	5	1	- <sup>b</sup>	- <sup>b</sup>	- <sup>b</sup>	18
Particle counting, particle/L	- <sup>b</sup>	6910	16450	5410	980	200	120	30	30100 <sup>c</sup>
Particle counting ratio <sup>b</sup> , %	- <sup>b</sup>	22.9	54.7	17.9	3.2	0.7	0.4	0.1	100 <sup>c</sup>
F.C. contribution <sup>c</sup> , MPN/L	$1.9 \times 10^6$	$4.5 \times 10^5$	$6.9 \times 10^5$	$2.4 \times 10^5$	$1.2 \times 10^4$	$4.2 \times 10^3$	$6.4 \times 10^2$	$1.6 \times 10^2$	$3.3 \times 10^6$
F.C. contribution, %	57.6	13.6	20.9	7.3	0.3	0.1	0.01	0.004	100

Notes: F.C. =  $3.3 \times 10^6$  MPN/L, SS = 18 mg/L,  $T = 25.5^\circ\text{C}$ ; a. Free F.C. included; b. the weight and number of particles could not be detected; c. the particles counting ratio is justly aim at particles with the size  $D > 2 \mu\text{m}$

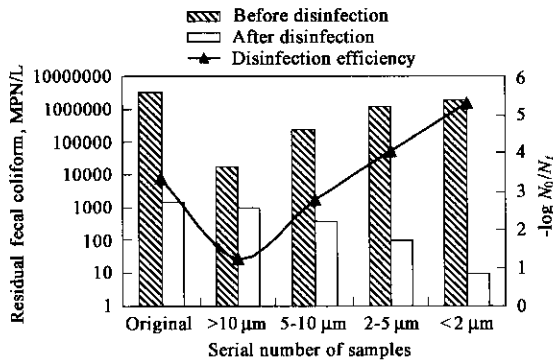


Fig.3 Effect of particle size on UV disinfection

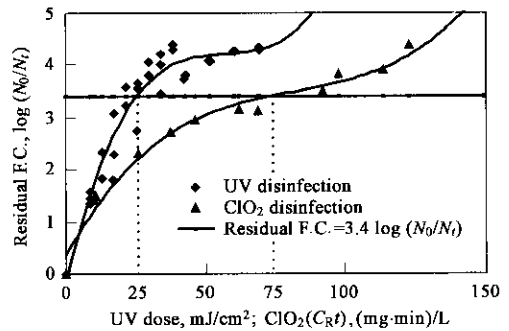
of effluent ( $D \geq 10 \mu\text{m}$ ) consistently produced a lower efficiency than the smaller fractions ( $D$ : 5—10  $\mu\text{m}$  and  $D$ : 2—5  $\mu\text{m}$ ), and the efficiency of three were 1.23, 2.78 and 4.1 log respectively. The free F.C. was the easiest to be disinfected, and the efficiency of which was more than 5.28 log. Although particles  $D \geq 10 \mu\text{m}$  associated F.C. were  $1.7 \times 10^4$ , about justly 0.5% of all F.C. of effluent (Table 2), this part F.C. was important for the tailings of F.C. inactivation curve. The result is close to the research of Emerick *et al.* (1999). Total F.C. after UV disinfection was  $1.5 \times 10^3$  MPN/L, and the F.C. from  $D \geq 10 \mu\text{m}$  particles was  $10^3$  MPN/L, about 66.7% of all. So for UV disinfection process, it is important to control particle concentration, especially  $D \geq 10 \mu\text{m}$  fraction.

Even through SS concentrations of different effluent samples were similar, the particle distribution of them will have significant differences because of the different treatment processes before disinfection. Therefore, the influence of particle size may explain why traditional solid measurements (total weight of suspended solid measurements) do not provide accurate prediction of UV disinfection. To ensure the disinfection efficiency in the UV disinfection, the particle counting apparatus is preferable to traditional solid measurement.

### 2.3 Dose of $\text{ClO}_2$ and UV process alone

Because the lifecycle in water and resistance of F.C. are similar to that of pathogenic intestinal bacteria, F.C. is commonly used to reflect the disinfection efficiency. The F.C. disinfection effects with respective  $\text{ClO}_2$  and UV are shown as Fig.4. The

disinfection standard is in reference to 1A class standard of Chinese National Integrated Wastewater Discharge Standard (GB81918-20020), which is 1000 MPN/L of F.C. The average of F.C. within effluent before disinfection was  $2.4 \times 10^6$ , the log inactivation was about 3.4 log to satisfy the level of 1000 MPN/L of F.C. Fig.4 shows that  $\text{ClO}_2$  disinfection  $C_{Rt}$  ( $t$  is contact time,  $C_R$  is residual  $\text{ClO}_2$  concentration after contract time) for this disinfection effect was  $75(\text{mg} \cdot \text{min})/\text{L}$ , and relative initial  $\text{ClO}_2$  dose was about 5 mg/L. And UV dose with the same disinfection effect was  $25 \text{ mJ}/\text{cm}^2$ .

Fig.4 Inactivation of F.C. with respective UV and  $\text{ClO}_2$  disinfection

### 2.4 Dose of $\text{ClO}_2$ /UV combined process

The  $\text{ClO}_2$  dose of  $\text{ClO}_2$ /UV disinfection was in reference to bio-toxicity after  $\text{ClO}_2$  disinfection (including  $\text{ClO}_2$  disinfection and then completely dechlorination, same in the following of this paper). The toxicity is shown in Fig.2. It can be seen that the bio-toxicity of wastewater at  $\text{ClO}_2$  dose of 4 mg/L was increased to 24%, which was close to the upper limit of low toxicity. Therefore,  $\text{ClO}_2$  dose of 3 mg/L was used in the  $\text{ClO}_2$ /UV disinfection process.

The samples were first disinfected with  $\text{ClO}_2$  for 30 min, then continued to be disinfected by different doses of UV. The total disinfection results are shown in Table 3. It can be found  $\text{ClO}_2$  and UV disinfection dose of 3 mg/L,  $6 \text{ mJ}/\text{cm}^2$  could meet the selected F.C. standard. Dechlorination was recommended before UV disinfection, not only to avoid the erosion of residual  $\text{ClO}_2$  to UV fittings, but to decrease the bio-toxicity of disinfected wastewater.

**Table 3 Inactivation of fecal coliform (F.C.) with ClO<sub>2</sub>/UV process**

ClO <sub>2</sub> dose, mg/L	UV dose, mJ/cm <sup>2</sup>	Residual F.C., MPN/L	Inactivation efficiency, -log(N <sub>t</sub> /N <sub>0</sub> )
0	0	2600000 (N <sub>0</sub> )	-
3	2	4600	2.78
3	4	1800	3.19
3	6	700	3.60
3	8	400	3.84
3	10	150	3.27
3	12	40	3.85

Notes: F.C.=2.6 × 10<sup>6</sup> MPN/L; SS=16 mg/L; T = 26.5°C

**2.5 Particle distribution after ClO<sub>2</sub> disinfection**

Both the particle size and weight distribution, at the end of 30 min contact after 3 mg/L ClO<sub>2</sub> disinfection, are shown in Table 4.

**Table 4 Particle distribution and weight after ClO<sub>2</sub> addition**

Particles size, μm	<2	2—3	3—5	5—10	10—20	20—30	30—50	>50	Total
SS distribution, mg/L	9	3	3	2	-	-	-	-	16
Particle counting, particle/L	-	10661	6182	2350	300	130	20	0	28770
Particle counting ratio, %	-	54.3	31.5	11.9	1.5	0.6	0.1	0	100

Notes: F.C.=3.3 × 10<sup>6</sup> MPN/L; SS=18 mg/L; T = 24.5°C

**2.6 Resistance to particle loadings of combined disinfection process**

A serial of samples with different particle concentrations before overflow weir of secondary clarifiers were selected, and SS effect on the three disinfection processes was studied and the result is shown in Fig.5. Although particle counting parameter was preferred to adjust the disinfection dose, SS parameter was used to reflect the total content of the particles within the effluent here. Fig.5 shows that UV disinfection efficiency decreased with the increase of SS, and both the disinfection efficiencies of ClO<sub>2</sub> and ClO<sub>2</sub>/UV processes were similar which had a little variation. It is obvious that the resistance ability to particle loadings of ClO<sub>2</sub>/UV disinfection process was better than UV process alone.

**2.7 Economic analysis**

The economic analysis of ClO<sub>2</sub> disinfection, UV, and combined ClO<sub>2</sub>/UV disinfection was calculated in Table 5. The disinfection objective was 1000 MPN/L, and disinfection doses were all from laboratory

**Table 5 Economic analysis of ClO<sub>2</sub>/UV disinfection process**

Process	Dose	Capital construction fund, USD	Operational fund, USD/a	Operational cost <sup>a</sup> , USD/m <sup>3</sup>
ClO <sub>2</sub>	5 mg/L	50620	294190	0.0110
UV	25 mJ/cm <sup>2</sup>	28100	18300	0.0005
ClO <sub>2</sub> /UV	3 mg L <sup>-1</sup> /6 mJ cm <sup>-2</sup>	28400/17800 <sup>b</sup>	177400/12800	0.0070

Notes: a. Dechlorination cost was not included in the capital construction fund, operational fund, but in the calculation of operational cost, experiential equation: S<sub>ClO<sub>2</sub> disinfection/dechlorination</sub> = 1.4 S<sub>ClO<sub>2</sub> disinfection</sub> was used to revise the results; b. the left and right of "/" were the respective cost for ClO<sub>2</sub> and UV of combined process

The difference between Table 2 and Table 4 shows that (1) SS concentration decreased slightly from 18 to 16 mg/L after chlorination; (2) the total number of particles D>2 μm was decreased from 30100 to 19643 particle/L; and the numbers of large particles decreased more evidently than the small ones. The total number of D>10 μm decreased by 65% (from 1330 to 450 particle/L), but the total number of particles D<10 μm decreased 33% (from 28770 to 19193 particle/L). The size of main particles of samples before chlorination is 3—5 μm, however that after chlorination is 2—3 μm. It can be explained that pre-ClO<sub>2</sub> addition can oxidize the organic matters within the particles, and made them split into small ones and then made the relatively smaller particles decomposed to dissolved state. As a result, both the total number of particles and the large ones descend, and the larger particles influencing on UV disinfection was reduced.

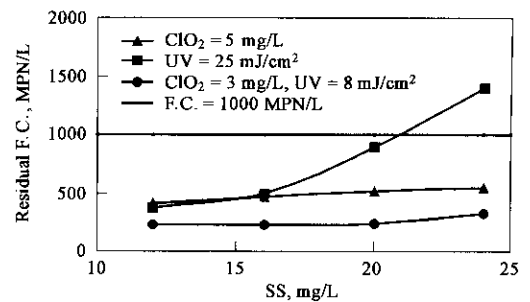


Fig.5 Effect of SS on three disinfection processes F.C.=3.4 × 10<sup>6</sup> MPN/L; T = 26.5°C

results. Capital cost includes the expense of equipments and relative buildings, and operational cost includes chemical disinfectants expense, manpower expense and equipment depreciation. The results show that the combined ClO<sub>2</sub>/UV disinfection process is more expensive than UV process alone, but inexpensive than ClO<sub>2</sub> disinfection with the same disinfection results.

### 3 Conclusions

Suspended solid size has a major impact on UV disinfection efficiency, and the large particle size fraction of effluent consistently produced a lower efficiency than the smaller ones. In order to ensure high disinfection efficiency, wastewater disinfected UV process should prefer particle counting apparatus to traditional solids measurements. Pre-ClO<sub>2</sub> addition oxidizes the organic matters within the particles, and make large particles decrease to small ones and then the relatively smaller particles are further transformed into dissolved state. As a result, the total number of particles in effluent decreases.

The combined ClO<sub>2</sub>/UV disinfection process has a higher resistance ability to particle loadings than UV process alone, and its bio-toxicity is much lower than ClO<sub>2</sub> process with same disinfection efficiency. Even though the operational cost of combined process is less than that of ClO<sub>2</sub> process. Furthermore, the combined process is easy to establish in the constructed and operated wastewater plants as well as those to be constructed and operated. For the wastewater plants that have constructed ClO<sub>2</sub> process, UV lamps can be installed at the end of dechlorination contact facilities, and need no other units. For a newly constructed wastewater treatment plant, the combined process requires less footprint and capital cost than that of traditional ClO<sub>2</sub> process, and has better disinfection efficiency than UV process.

### References:

- Andreadakis A, Mamais D, Christoulas D *et al.*, 1999. Ultraviolet disinfection of secondary and tertiary effluent in the Mediterranean region[J]. *Water Sci Technol*, 40(4/5): 253—260.
- APHA, 1998. Standard methods for the examination of water and wastewater. [M]. 20th ed. Washington: American Public Health Association.
- Berman D R, Eugene W, Hoff J C, 1988. Inactivation of particle-associated coliforms by chlorine and monochloramine[J]. *Applied Environ Microbiol*, 54(2): 507—512.
- Cheng L H, Gao H W, Ni F X, 2003. The use of ultraviolet disinfection in sewage treatment [J]. *Techniques and Equipment for Environmental Pollution Control*, 12(4): 69—72.
- Chiu K, Lun D A, Savoye P *et al.*, 1999. Integrated UV disinfection model based on particle trickling [J]. *J Am Water Works Assoc*, 125(1): 7—16.
- Emerick R W, Loge F, Thompson D *et al.*, 1999. Factors influencing ultraviolet disinfection performance. part 2: Association of coliform bacteria with wastewater particles [J]. *Water Environ Res*, 71(6): 1178—1187.
- Frank J L, Robert W E, Tim R G *et al.*, 2002. Association of coliform bacteria with wastewater particles: impact of operational parameters of the activated sludge process [J]. *Water Res*, 36: 41—48.
- George T, Franklin L B, 2003. Wastewater engineering treatment and reuse[M]. 4th ed. Washington: Metcalf & Eddy Inc. 653—702.
- John E T, Ernest R B, 1999. Toxicity effects of  $\gamma$ -irradiated wastewater effluents[J]. *Water Res*, 33(9): 2053—2058.
- Kaiser K, Palabrica S, 1991. Photobacterium phosphoreum toxicity data index[J]. *Water Pollut Res*, 26(3): 361—431.
- Ronald G, Monika W, Priya V *et al.*, 2003. Disinfection efficiency of peracetic acid, UV and ozone after enhanced primary treatment of municipal wastewater[J]. *Water Res*, 37: 4573—4586.
- White G C, 1999. The handbook of chlorination and alternative disinfectants [M]. 4th ed. New York: John Wiley and Sons Inc., 4.4: 1153—1202.

(Received for review September 26, 2005. Accepted November 21, 2005)