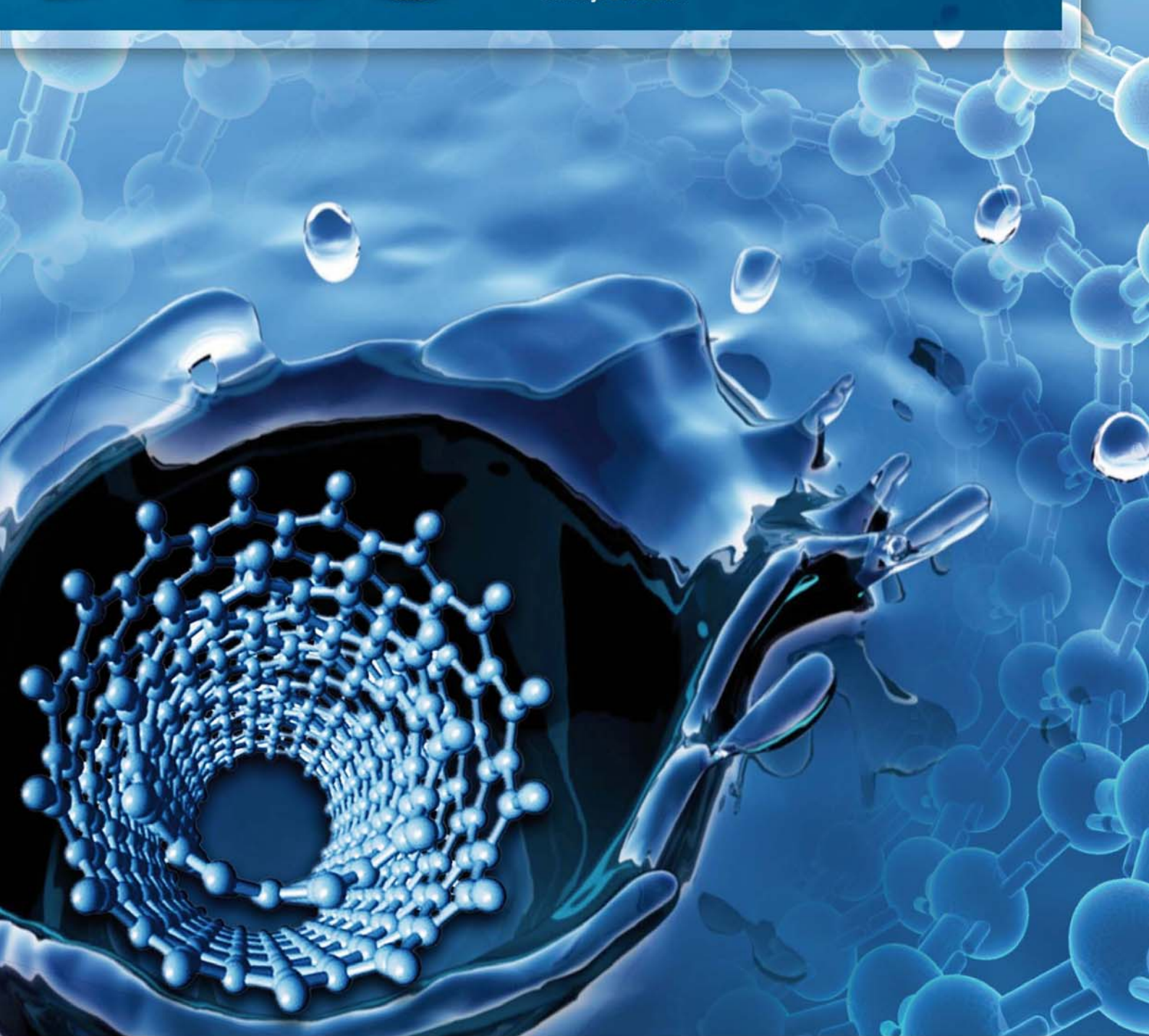


# JES

JOURNAL OF  
ENVIRONMENTAL  
SCIENCES

ISSN 1001-0742  
CN 11-2629/X

July 1, 2013 Volume 25 Number 7  
[www.jesc.ac.cn](http://www.jesc.ac.cn)



Sponsored by  
Research Center for Eco-Environmental Sciences  
Chinese Academy of Sciences

## CONTENTS

### Aquatic environment

- Application potential of carbon nanotubes in water treatment: A review  
Xitong Liu, Mengshu Wang, Shujuan Zhang, Bingcai Pan ..... 1263
- Characterization, treatment and releases of PBDEs and PAHs in a typical municipal sewage treatment plant situated beside an urban river, East China  
Xiaowei Wang, Beidou Xi, Shouliang Huo, Wenjun Sun, Hongwei Pan, Jingtian Zhang, Yuqing Ren, Hongliang Liu ..... 1281
- Factors influencing antibiotics adsorption onto engineered adsorbents  
Mingfang Xia, Aimin Li, Zhaolian Zhu, Qin Zhou, Weiben Yang ..... 1291
- Assessment of heavy metal enrichment and its human impact in lacustrine sediments from four lakes in the mid-low reaches of the Yangtze River, China  
Haijian Bing, Yanhong Wu, Enfeng Liu, Xiangdong Yang ..... 1300
- Biodegradation of 2-methylquinoline by *Enterobacter aerogenes* TJ-D isolated from activated sludge  
Lin Wang, Yongmei Li, Jingyuan Duan ..... 1310
- Inactivation, reactivation and regrowth of indigenous bacteria in reclaimed water after chlorine disinfection of a municipal wastewater treatment plant  
Dan Li, Siyu Zeng, April Z. Gu, Miao He, Hanchang Shi ..... 1319
- Photochemical degradation of nonylphenol in aqueous solution: The impact of pH and hydroxyl radical promoters  
Aleksandr Dulov, Niina Dulova, Marina Trapido ..... 1326
- A pilot-scale study of cryolite precipitation from high fluoride-containing wastewater in a reaction-separation integrated reactor  
Ke Jiang, Kanggen Zhou, Youcai Yang, Hu Du ..... 1331

### Atmospheric environment

- Effect of phosphogypsum and dicyandiamide as additives on NH<sub>3</sub>, N<sub>2</sub>O and CH<sub>4</sub> emissions during composting  
Yiming Luo, Guoxue Li, Wenhai Luo, Frank Schuchardt, Tao Jiang, Degang Xu ..... 1338
- Evaluation of heavy metal contamination hazards in nuisance dust particles, in Kurdistan Province, western Iran  
Reza Bashiri Khuzestani, Bubak Sourì ..... 1346

### Terrestrial environment

- Utilizing surfactants to control the sorption, desorption, and biodegradation of phenanthrene in soil-water system  
Haiwei Jin, Wenjun Zhou, Lizhong Zhu ..... 1355
- Detoxifying PCDD/Fs and heavy metals in fly ash from medical waste incinerators with a DC double arc plasma torch  
Xinchao Pan, Jianhua Yan, Zhengmiao Xie ..... 1362
- Role of sorbent surface functionalities and microporosity in 2,2',4,4'-tetrabromodiphenyl ether sorption onto biochars  
Jia Xin, Ruilong Liu, Hubo Fan, Meilan Wang, Miao Li, Xiang Liu ..... 1368

### Environmental biology

- Systematic analysis of microfauna indicator values for treatment performance in a full-scale municipal wastewater treatment plant  
Bo Hu, Rong Qi, Min Yang ..... 1379
- Function of *arsATorf7orf8* of *Bacillus* sp. CDB3 in arsenic resistance  
Wei Zheng, James Scifleet, Xuefei Yu, Tingbo Jiang, Ren Zhang ..... 1386
- Enrichment, isolation and identification of sulfur-oxidizing bacteria from sulfide removing bioreactor  
Jianfei Luo, Guoliang Tian, Weitie Lin ..... 1393

---

## Environmental health and toxicology

- In vitro* immunotoxicity of untreated and treated urban wastewaters using various treatment processes to rainbow trout leucocytes  
François Gagné, Marlène Fortier, Michel Fournier, Shirley-Anne Smyth ..... 1400
- Using lysosomal membrane stability of haemocytes in *Ruditapes philippinarum* as a biomarker of cellular stress  
to assess contamination by caffeine, ibuprofen, carbamazepine and novobiocin  
Gabriela V. Aguirre-Martínez, Sara Buratti, Elena Fabbri, Angel T. DelValls, M. Laura Martín-Díaz ..... 1408

## Environmental catalysis and materials

- Effect of transition metal doping under reducing calcination atmosphere on photocatalytic  
property of TiO<sub>2</sub> immobilized on SiO<sub>2</sub> beads  
Rumi Chand, Eiko Obuchi, Katsumi Katoh, Hom Nath Luitel, Katsuyuki Nakano ..... 1419
- A high activity of Ti/SnO<sub>2</sub>-Sb electrode in the electrochemical degradation of 2,4-dichlorophenol in aqueous solution  
Junfeng Niu, Dusmant Maharana, Jiale Xu, Zhen Chai, Yueping Bao ..... 1424
- Effects of rhamnolipid biosurfactant JBR425 and synthetic surfactant Surfynol465 on the  
peroxidase-catalyzed oxidation of 2-naphthol  
Ivanec-Goranina Rūta, Kulys Juozas ..... 1431

## The 8th International Conference on Sustainable Water Environment

- An novel identification method of the environmental risk sources for surface water pollution accidents in chemical industrial parks  
Jianfeng Peng, Yonghui Song, Peng Yuan, Shuhu Xiao, Lu Han ..... 1441
- Distribution and contamination status of chromium in surface sediments of northern Kaohsiung Harbor, Taiwan  
Cheng-Di Dong, Chiu-Wen Chen, Chih-Feng Chen ..... 1450
- Historical trends in the anthropogenic heavy metal levels in the tidal flat sediments of Lianyungang, China  
Rui Zhang, Fan Zhang, Yingjun Ding, Jinrong Gao, Jing Chen, Li Zhou ..... 1458
- Heterogeneous Fenton degradation of azo dyes catalyzed by modified polyacrylonitrile fiber Fe complexes:  
QSPR (quantitative structure property relationship) study  
Bing Li, Yongchun Dong, Zhizhong Ding ..... 1469
- Rehabilitation and improvement of Guilin urban water environment: Function-oriented management  
Yuansheng Pei, Hua Zuo, Zhaokun Luan, Sijia Gao ..... 1477
- Adsorption of Mn<sup>2+</sup> from aqueous solution using Fe and Mn oxide-coated sand  
Chi-Chuan Kan, Mannie C Aganon, Cybelle Morales Futralan, Maria Lourdes P Dalida ..... 1483
- Degradation kinetics and mechanism of trace nitrobenzene by granular activated carbon enhanced  
microwave/hydrogen peroxide system  
Dina Tan, Honghu Zeng, Jie Liu, Xiaozhang Yu, Yanpeng Liang, Lanjing Lu ..... 1492

Serial parameter: CN 11-2629/X\*1989\*m\*237\*en\*P\*28\*2013-7



## Historical trends in the anthropogenic heavy metal levels in the tidal flat sediments of Lianyungang, China

Rui Zhang<sup>1,2,\*</sup>, Fan Zhang<sup>3</sup>, Yingjun Ding<sup>1</sup>, Jinrong Gao<sup>1</sup>,  
Jing Chen<sup>1</sup>, Li Zhou<sup>1,\*</sup>

1. School of Geodesy and Geomatics Engineering, Huaihai Institute of Technology, Lianyungang, Jiangsu 222005, China

2. State Key Laboratory of Pollution Control and Resources Reuse, Nanjing University, Nanjing 210093, China

3. Jiangsu Marine Resources Development Research Institute, Huaihai Institute of Technology, Lianyungang, Jiangsu 222001, China

### Abstract

The sedimentation of metals can preserve the historical record of contaminant input from local and regional sources and provide information on the historical changes in regional water and sediment quality. We report the <sup>210</sup>Pb activities and the heavy metal (Cd, Cr, Cu, Mn, Pb and Zn) depth profiles from sediment cores retrieved in 2010. The mean sedimentation rates of 0.85–1.5 cm/yr are determined by <sup>210</sup>Pb dating. The sediments in the tidal flat have recorded heavy metal deposition and thus allow the establishment of a connection between the temporal evolution of the heavy metal pollution and the historical changes in the economic development of Lianyungang. The enrichment factors (EF) are calculated to estimate the level of contamination stored in these sediments. The results show that in the studied sites, Cr and Cu display low EF values and are mainly from lithogenic origin. For the other studied trace metals, a great variability in the sedimentary record is observed. Significant anthropogenic enrichment over the last 50 years is revealed at the tidal flat that receives fluvial inputs. Zinc is the element with the highest EF values, followed by the order of Pb > Cd > Mn > Cu and Cr. The temporal variations of the heavy metals peak during the late 1980s to the early 2000s and show a decreasing trend afterward. The pollution intensity of the tidal flat is determined by using EF and the geo-accumulation index ( $I_{geo}$ ), which show that, based on the  $I_{geo}$  scale, the tidal flat of Haizhou Bay is unpolluted to moderately polluted.

**Key words:** sediment rates; heavy metals; pollution; anthropogenic input; tidal flat

**DOI:** 10.1016/S1001-0742(12)60186-7

### Introduction

Due to their toxicity, bioaccumulation capacity and persistence, the cycling of heavy metals is a serious concern that has been the focus of many studies of coastal wetland environments (Eisenrhcil et al., 1986; Lacerda et al., 1988; Benoit et al., 1999; Li et al., 2007; Niu et al., 2009). Coastal tidal marshes play critical buffering roles in slowing shoreline erosion and in absorbing nutrients and contaminants before they reach the oceans and estuaries, which can result in the accumulation of excess levels of contaminants in marshes near urban and industrial areas. As a result of rapid industrial development over the past century, heavy metals have been discharged into coastal environments (Lee and Cundy, 2001; Bellucci et al., 2002; Munksgaard et al., 2003; Spencer et al., 2003; Maanan, 2008). Increased urbanization and industrialization have led to increased marine discharges, in turn, the total load of pollutants,

including heavy metals, being delivered to the sea has increased (Natesan et al., 2010). Because of their variable physical and chemical properties, coastal zones may act as sinks or sources of heavy metals (Harbison, 1986). The heavy metals discharged into water are generally adsorbed onto suspended particles and then deposited on the bottom as sediments. Consequently, the heavy metals in sediment cores can be used to reconstruct historical fluctuations in regional water and sediment quality (Hartmann et al., 2005; Cantwell et al., 2007). With this information, the efficiency of pollution management strategies can be evaluated, and regulatory actions can be implemented to reduce potential health risks.

Haizhou Bay lies on the western margin of the South Yellow Sea, near the city of Lianyungang, and receives water from the Linhong River (**Fig. 1**). The bay, which is shaped like a trumpet, has an area of approximately  $876.39 \times 10^6$  m<sup>2</sup>, a coastline length of  $86.81 \times 10^3$  m, and a maximum width of approximately  $42 \times 10^3$  m. Over the last 80 years, the bay has been affected by anthropogenic

\* Corresponding author. E-mail: [rzhang\\_838@163.com](mailto:rzhang_838@163.com) (Rui Zhang); [zhoulilyg@sina.com](mailto:zhoulilyg@sina.com) (Li Zhou)

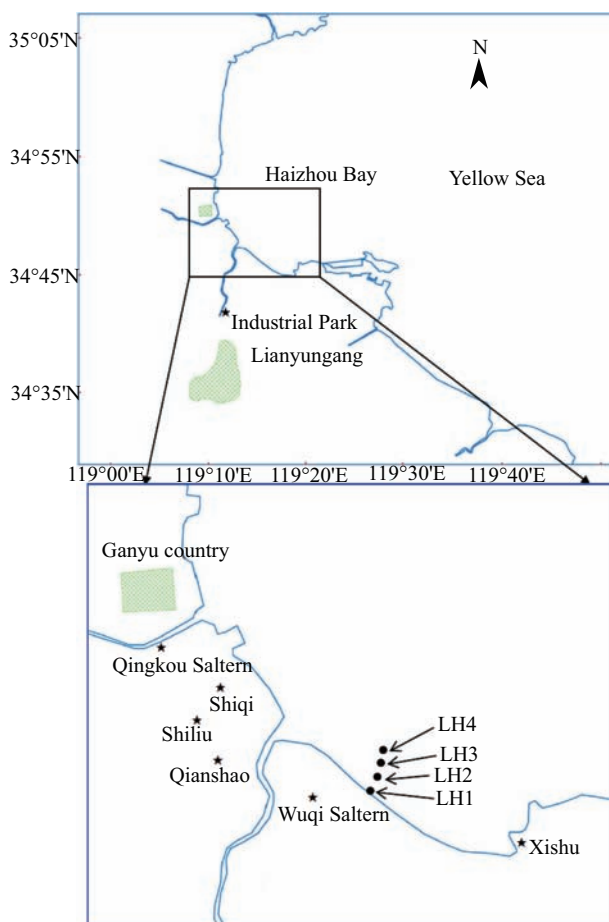


Fig. 1 Sampling site map.

activities in the surrounding area, including reclamation, the construction of harbors and dikes, the dredging of a channel, maritime transport and aquaculture (Chen, 2006). However, the urban infrastructure needed to support the population has not been developed, and substantial amounts of sewage are discharged into the sea or the Linhong River without treatment.

For more than 40 years, various industrial activities near Haizhou Bay have significantly contaminated the coastal tidal flats. A few studies have reported the geochemical composition and the characteristics of both major and trace elements in sediment cores (Zhang et al., 2008). However, there are limited data on historical toxic metals inputs to the tidal flats of Haizhou Bay. This gap needs to be filled because this information will help to predict future trends in the marsh sediment quality and to decide if further action is required to accelerate the recovery of the ecosystem in Haizhou Bay. Moreover, the levels and the orientation of industrialized development near Haizhou Bay in Lianyungang provide an interesting contrast for understanding pollutant sources and distribution mechanisms in similar aquatic environment. This work aims to reconstruct historical changes in terms of heavy metal (Cd, Cr, Cu, Mn, Pb, and Zn) pollution in the tidal flat of Haizhou Bay.

## 1 Materials and methods

### 1.1 Sampling and chemical analysis

Four sediment cores were manually collected from the south tidal flat of Haizhou Bay, near the Linhong River estuary, in April 2010. The cores were labeled LH1 (34.774°N, 119.247°E), LH2 (34.778°N, 119.249°E), LH3 (34.782°N, 119.251°E) and LH4 (34.787°N, 119.252°E), and their locations are shown in Fig. 1. In the laboratory, these cores, which ranged from 12 to 65 cm in length and which were 7.5 cm in diameter, were subsampled at 1 cm intervals. Each layer was divided for analysis of the metals, the grain size and the  $^{210}\text{Pb}$  radionuclide.

The analytical procedures for the measurement of metal concentrations in core sediments are similar to those for soils and sediments. The samples were completely digested by a micro-wave assisted digestion method (1200 W, 180°C, 1 MPa during 30 min) with concentrated HF, HNO<sub>3</sub> and HCl acids in Teflon bottle, and then the samples were placed on a hot plate to evaporated until dry. The residue was solubilised by 5% HNO<sub>3</sub> and diluted to volume. The metal concentrations were determined by an iCAP 6300 inductively coupled plasma optical emission spectrometer (ICP-OES, Thermo Electro Co., USA). The accuracy of the analytical method was established with the standard reference material aquatic sediments GBW07312 (GSD-12). Based on the analysis of the standard reference material, the recovery for each of the analyzed elements were as follows: Al (97%–105%), Cd (91%–110%), Cr (95%–103%), Cu (95%–105%), Mn (98%–106%), Pb (93%–108%) and Zn (99%–105%). Three replicates of each sample were run, and the following coefficients of variation were obtained: Al, 0.44%; Cd, 7.02%; Cr, 5.97%; Cu, 0.95%; Mn, 0.60%; Pb, 3.04%; and Zn, 2.22%.

In addition, a grain size analysis was performed on a part of each homogenized sample using a laser particle size analyzer (Mastersize 2000, Malvern Instruments Ltd., UK), after the removal of carbonate with 1 mol/L HCl and of organic matter using 30% H<sub>2</sub>O<sub>2</sub>.

The activities of  $^{210}\text{Pb}$  were determined by alpha counting of  $^{210}\text{Po}$  deposited onto Ag discs (Flynn, 1968) using  $^{209}\text{Po}$  as a yield tracer. The  $^{209}\text{Po}$  spike had previously been calibrated against the standard reference material (IAEA-257). The sediment was dissolved by adding a solution of 1:1:0.5 (V/V/V) HNO<sub>3</sub>+HCl+HF to 2 g sediment and heating the mixture to 200°C overnight in a closed Teflon bottle. Counting was conducted by computerized multi-channel  $\alpha$ -spectrometry with gold-silicon surface barrier detectors (Model Octete PLUS, ORTEC Co., USA).

### 1.2 Statistic analysis

Statistic analysis was carried out using Microsoft Office Excel and the Statistical Package for Social Science (SPSS 16.0 for Windows, SPSS Inc., USA). Nonparametric test was used to determine any significant difference in metal

concentrations among different sample sites at the level of  $p < 0.05$ . Correlations among metal concentrations in the water, soils and plants samples were evaluated using Pearson correlation coefficients.

## 2 Results and discussion

### 2.1 Sediment characteristics

The sediments in cores LH3 and LH4 are primarily yellow-brown mud; and then towards the bottom of the cores, the sediments change to a dark grey mud with a few black stripes that contains some roots and some shell fragments. The sediments of cores LH1 and LH2 are brown and dark brown without any observable change in the visual or physical structures. The four cores consist of relatively uniform fine-grained materials, with the percentage of silt (4–63  $\mu\text{m}$ ) and clay (< 4  $\mu\text{m}$ ) exceeding 95% (Fig. 2). The silt fraction ranges from 62.5% to 95.9%, and the clay

fraction ranges from 2.0% to 27.7%. The percentage of sand is very small in these cores.

### 2.2 Sedimentation rate and age of cores LH3 and LH4

Figure 3 shows the vertical distributions of the excess  $^{210}\text{Pb}$  for the cores LH3 and LH4. The supported  $^{210}\text{Pb}$  activity (derived from the decay of its effective parent,  $^{226}\text{Ra}$ , in the seabed) is generally determined by measuring the  $^{210}\text{Pb}$  activity at the bottom of a core that exhibits low accumulation rates, such that sediment is greater than 100 years old. However, the bottom layers of the four cores do not approach the depth of supported  $^{210}\text{Pb}$  activity, thus the supported  $^{210}\text{Pb}$  activity was approximated to be 1 dpm/g (decays per minute per gram), as taken from previous studies (Zhang et al., 2008; Chen, 2006). The excess  $^{210}\text{Pb}$  activity ( $^{210}\text{Pb}_{\text{ex}}$ , the activity particles attain while sinking through the water column) was calculated by subtracting the supported activity from the total activity. The excess

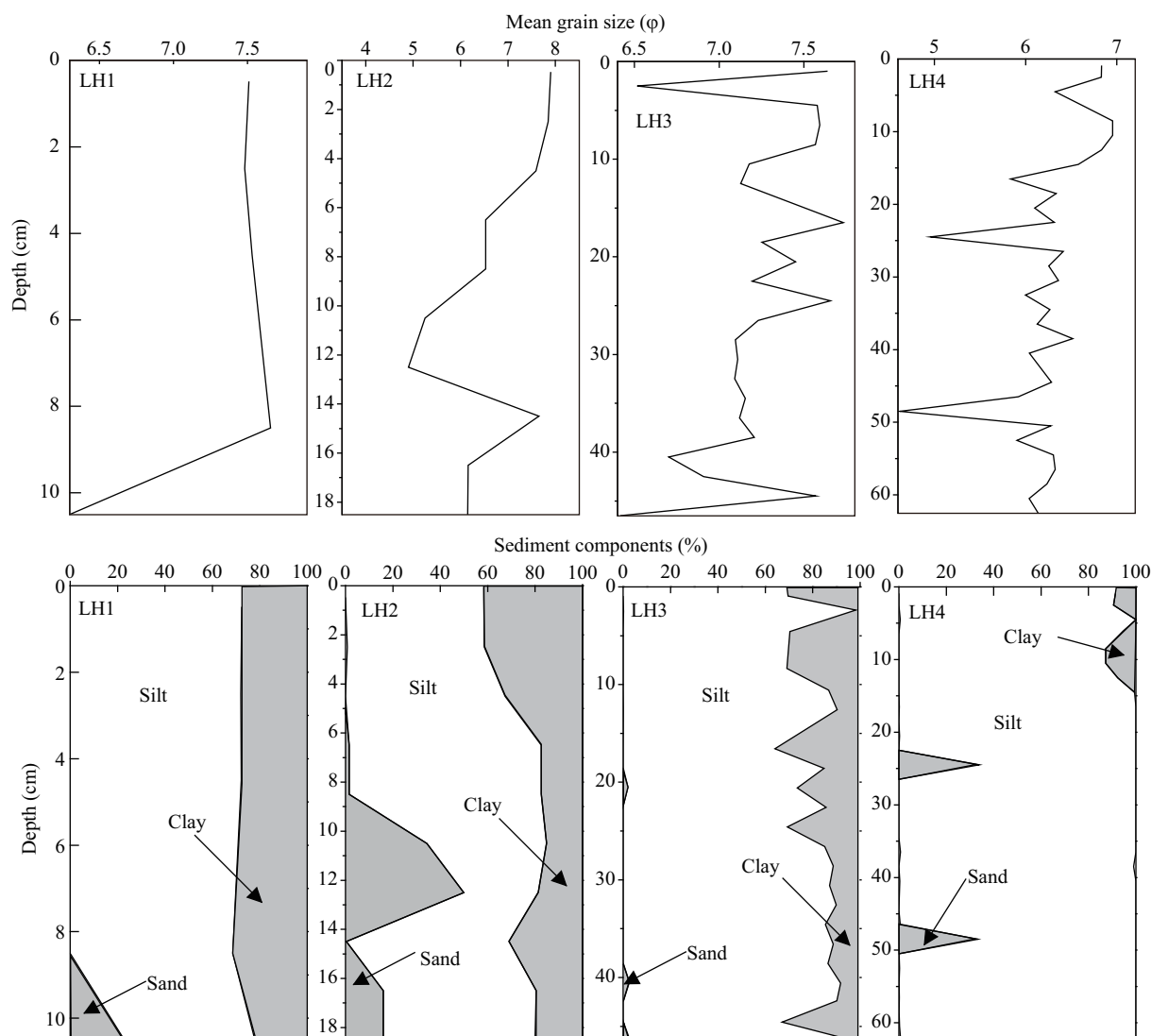
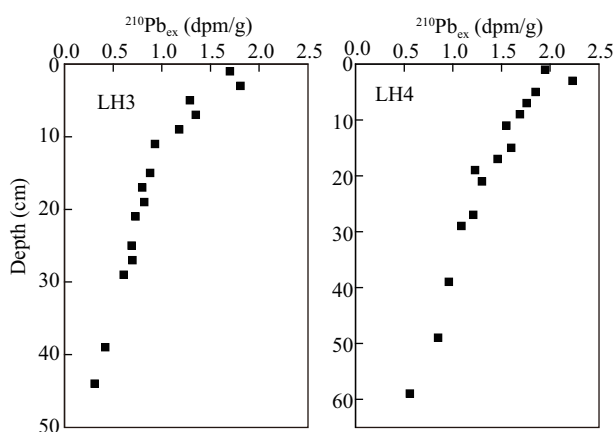


Fig. 2 Profiles of mean grain size and sediment components in sediment cores from tidal flat in Haizhou Bay.  $\phi = -\log_2 d$ , where,  $d$  (mm) is the diameter of grain.



**Fig. 3** Depth profiles of excess  $^{210}\text{Pb}$  ( $^{210}\text{Pb}_{\text{ex}}$ ) of cores LH3 and LH4, an exponential decreasing trend with depth of excess  $^{210}\text{Pb}$ . dpm/g: decays per minute per gram

$^{210}\text{Pb}$  activities exhibit an approximately exponentially decreasing trend with depth (Fig. 3), which suggest a constant deposition of excess  $^{210}\text{Pb}$  over time at these sites. On the basis of the  $^{210}\text{Pb}_{\text{ex}}$  activities, the constant initial concentration (CIC) model was used to calculate sedimentation rates. The CIC model has been successfully applied in other studies of estuaries and intertidal mudflats (Andersen et al., 2000; Sanders et al., 2010). For this model, the  $^{210}\text{Pb}_{\text{ex}}$  activity ( $A_z$ ) at any sediment layer ( $z$ ) with age ( $t$ ) is expressed as:

$$A_z = A_0 \times e^{-\lambda t} = A_0 \times e^{-\lambda z/S} \quad (1)$$

where,  $A_0$  (dpm/g) is the  $^{210}\text{Pb}_{\text{ex}}$  at the sediment-water interface,  $\lambda$  is the radioactive decay constant for  $^{210}\text{Pb}$  ( $0.03114 \text{ yr}^{-1}$ ), and  $S$  (cm/yr) is the sedimentation rate. From Eq. (1),  $S$  was determined by the slope of the  $^{210}\text{Pb}_{\text{ex}}$  profiles using least squares regression. However, a number of studies have shown that the compaction on sediment layers may cause an incorrect depth ( $z$ ) (Lu, 2007). An alternative method, which expresses  $^{210}\text{Pb}_{\text{ex}}$  as a function of massdepth removes this compaction effect (Lu, 2007). Thus, Eq. (1) can be rewritten as:

$$A_m = A_0 \times e^{-\lambda t} = A_0 \times e^{-\lambda m/r} \quad (2)$$

where,  $m$  ( $\text{g}/\text{cm}^2$ ) is the mass depth of the cumulative dry weight at the sediment layer ( $z$ ),  $r$  ( $\text{g}/(\text{cm}^2 \cdot \text{yr})$ ) is the sediment accumulation rate,  $A_m$  (dpm/g) is the  $^{210}\text{Pb}_{\text{ex}}$  at the sediment layer ( $m$ ), and  $A_0$  is the  $^{210}\text{Pb}_{\text{ex}}$  at the sediment-water interface layer. From Eq. (2),  $r$  was determined by the slope of the  $^{210}\text{Pb}_{\text{ex}}$  profiles using least squares regression.

Finally, the  $^{210}\text{Pb}$  chronologies of the two methods were compared to derive the final chronology. The results by the two methods agreed well, it means that the compaction did not have a significant effect on the  $^{210}\text{Pb}$  chronologies in the two sediment cores. Therefore, the result of the sediment chronology was unique and was reported by the least

squares regression from Eq. (1). The mean sedimentation rate was calculated to be 1.5 cm/yr for core LH4, and 0.85 cm/yr for core LH3. The sediments in these cores have accumulated over approximately the last 50 years at a constant sedimentation rate. The deepest layer in core LH4, at a depth of 63 cm, was estimated to have been deposited in 1965. For core LH3, the layer at a depth of 47 cm corresponds to 1950. Therefore, these cores should contain records related to the anthropogenic impacts of the pollutants from Lianyungang City. There have been several reports of  $^{210}\text{Pb}$  dating of cores from Haizhou Bay. One core from the northern area near the estuary of the Linhong River was determined to have a sedimentation rate of 0.5 cm/yr (Zhang et al., 2008), while another core near Xishu Cape had a rate of 1.15 cm/yr (Peng and Chen, 2010). The variability of the three cores reflects the fact that the sedimentation rate is relatively high in the southern area of Haizhou Bay and decreases gradually moving northwards. According to a previous study (Wang et al., 1980), the southern tidal flat of Haizhou Bay is a rapid sedimentation area with a high average sedimentation rate between 1.0 and 2.5 cm/yr, which is due to the sediment loads from the Linhong River, as well as the abundant sediment from the Old Yellow Delta carried by northward coastal currents.

### 2.3 Metal contents in the deposits

Figure 4 shows the vertical distribution of heavy metals in the cores. The concentration of these elements in the deepest layer of core LH4 is almost equal to that of sediment without anthropogenic contamination. Except for core LH1, the heavy metal concentrations increased moving from the bottom to the top, and the maximum values were observed at depths of 5–20 cm. In the upper layer, the concentrations decreased at the surface. The vertical profiles of the heavy metals reflect the historical trend of the load produced by human activity in the coastal area of Haizhou Bay, as will be discussed in the following sections.

The correlation matrices for samples were calculated to examine the relationships between the elements analyzed (Table 1). Anthropogenic Cd, Cr, Cu, Mn and Zn show statistically significant ( $p < 0.05$ ) correlation among them. They most likely have the same origin and are influenced by the same mechanisms and processes. The main sources of heavy metals in soils from anthropogenic activities are atmospheric deposition, the waste from industrial and domestic activities and traffic-related emissions in urban areas (Zourarah et al., 2007; Liu et al., 2003). There are positive correlations between Pb and the other elements, with coefficients ranging from 0.34 to 0.46. These relationships indicate that Pb originates from a different source than the other elements. Anthropogenic Pb comes mainly from non-point sources such as atmospheric transport and precipitation, which incorporate Pb that originated from coal burning in power generation plants and other industri-

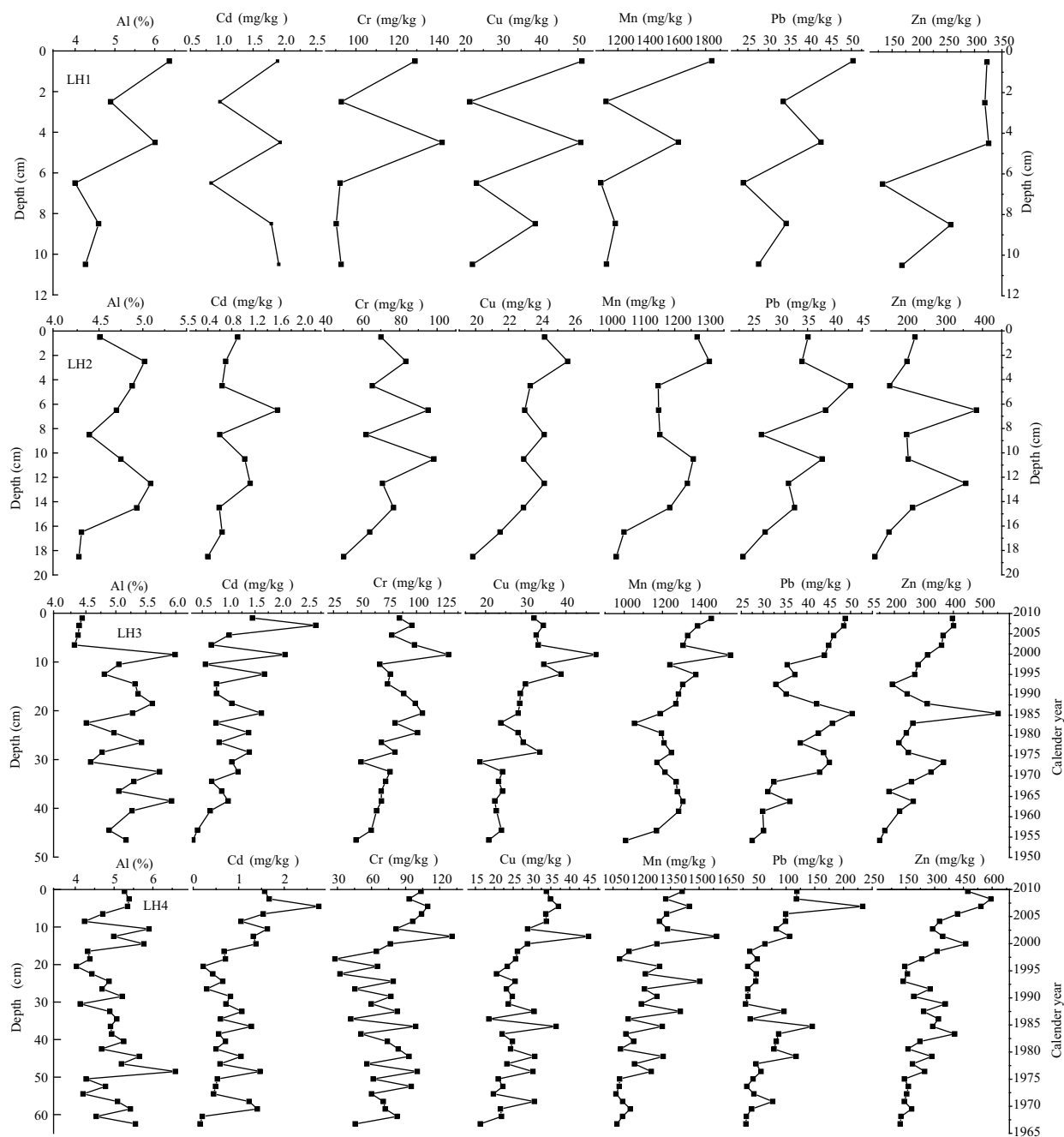


Fig. 4 Profiles of elements in sediment cores from tidal flat in Haizhou Bay.

al activities (Li et al., 2000; Liu et al., 2003; Ip et al., 2004). It is well known that trace metals tend to be adsorbed onto the surfaces of particles and that higher concentrations of metals are usually associated with smaller particles (Schiff and Weisberg, 1999; Tang et al., 2008). In the tidal flat of Haizhou Bay, the trace metals, including Cd, Cr, Cu, Mn, Pb and Zn, showed no significant correlation with the percentage of fine-grained particles in the sediments ( $p > 0.05$ ) (Table 1). Therefore, anthropogenic inputs were likely a more significant control on the distribution of trace metals in the tidal flat sediments than natural geochemical processes.

#### 2.4 Enrichment of anthropogenic heavy metals

Since the 1970s, the tidal flats have been at risk of heavy metal pollution from industrial and agricultural discharge and from domestic wastewater due to rapid economic development in Lianyungang City near Haizhou Bay. In previous studies, the heavy metals contamination of the Haizhou Bay sediments has been demonstrated by increases in their concentrations (Zhang et al., 2008). The possibility of sediment contamination was assessed through the calculation of the metal enrichment factor (EF), which is the observed metal/normalizer ratio in the sample divided by the metal/normalizer ratio reported for



**Table 1** Comparisons of concentrations of metals (mg/kg) in surface and cores in tidal flat from Haizhou Bay with background values

		Cd	Cr	Cu	Mn	Pb	Zn
Surface	LH1	1.89	129.18	51.07	1841.84	50.41	322.76
	LH2	0.90	70.10	24.20	1268.59	35.08	225
	LH3	1.45	84.63	32.04	1455.72	48.81	396.97
	LH4	1.57	103.58	33.82	1411.27	118.82	472.75
Average in sediment cores	LH1	1.55	107	34.6	1327	35.4	254
	LH2	0.83	73.66	23.17	1177	32.9	224.06
	LH3	1.08	80	28.8	1260	39.4	284
	LH4	0.93	75.4	27.5	1246	73	286
CS <sup>a</sup>		0.365	60.28	15.84	– <sup>e</sup>	24.7	64.68
SYS <sup>b</sup>		0.088	64	18	570	22	67
MS <sup>c</sup>		0.06	15	8	– <sup>e</sup>	21	51
PB <sup>d</sup>		1.0	90	50	– <sup>e</sup>	70	175

<sup>a</sup> Geochemical natural background of coastal soil in Jiangsu (Xia et al., 1987); <sup>b</sup> background of sediments in Yellow Sea (Chi and Yan, 2007); <sup>c</sup> background of marine sediments in Hong Kong (EPD, 1992); <sup>d</sup> natural, preindustrial background values (Hakanson, 1980); <sup>e</sup> no data.

a reference material (Ip et al., 2004; Li et al., 2010). The EF was calculated as the ratio of the elemental concentration in the sediment to its natural background concentration (Tanner et al., 2000; Zhang and Liu, 2002; Feng et al., 2004; Selvaraj et al., 2004; Shi et al., 2010). A reference element was employed in the calculation to account for the lithogenic influence of granulometric and mineralogical variations of the sediment. Most commonly, Fe and Al are used (Lalraj and Nair, 2006; Aksu et al., 2010). The EF in this study is defined as

$$EF = (C_{M-sed}/C_{R-sed}) / (C_{M-bk}/C_{R-bk}) \quad (3)$$

where,  $(C_{M-sed}/C_{R-sed})$  is the ratio of the heavy metal concentration ( $C_{M-sed}$ ) to concentration of the reference element ( $C_{R-sed}$ ) in the sediment, and  $(C_{M-bk}/C_{R-bk})$  is the concentration ratio for the natural background. The natural geochemical background refers to the elemental concentrations in sediments without any anthropogenic enrichment; i.e., the elemental concentrations that are fully derived from natural sources such as terrigenous and biogenic sources (Rubio et al., 2000; Liaghati et al., 2003; Cobelo-Garcia and Prego, 2003; Mil-Homens et al., 2006; Shi et al., 2010). In previous studies, the natural geochemical background has not been extensively reported for Haizhou Bay, because a short core is not sufficient to reconstruct a geochemical history unaffected by any human influence. Therefore, the natural geochemical background of the coastal soil in Jiangsu (Xia et al., 1987) was taken as a

background value for Haizhou Bay and is listed in **Table 2**.

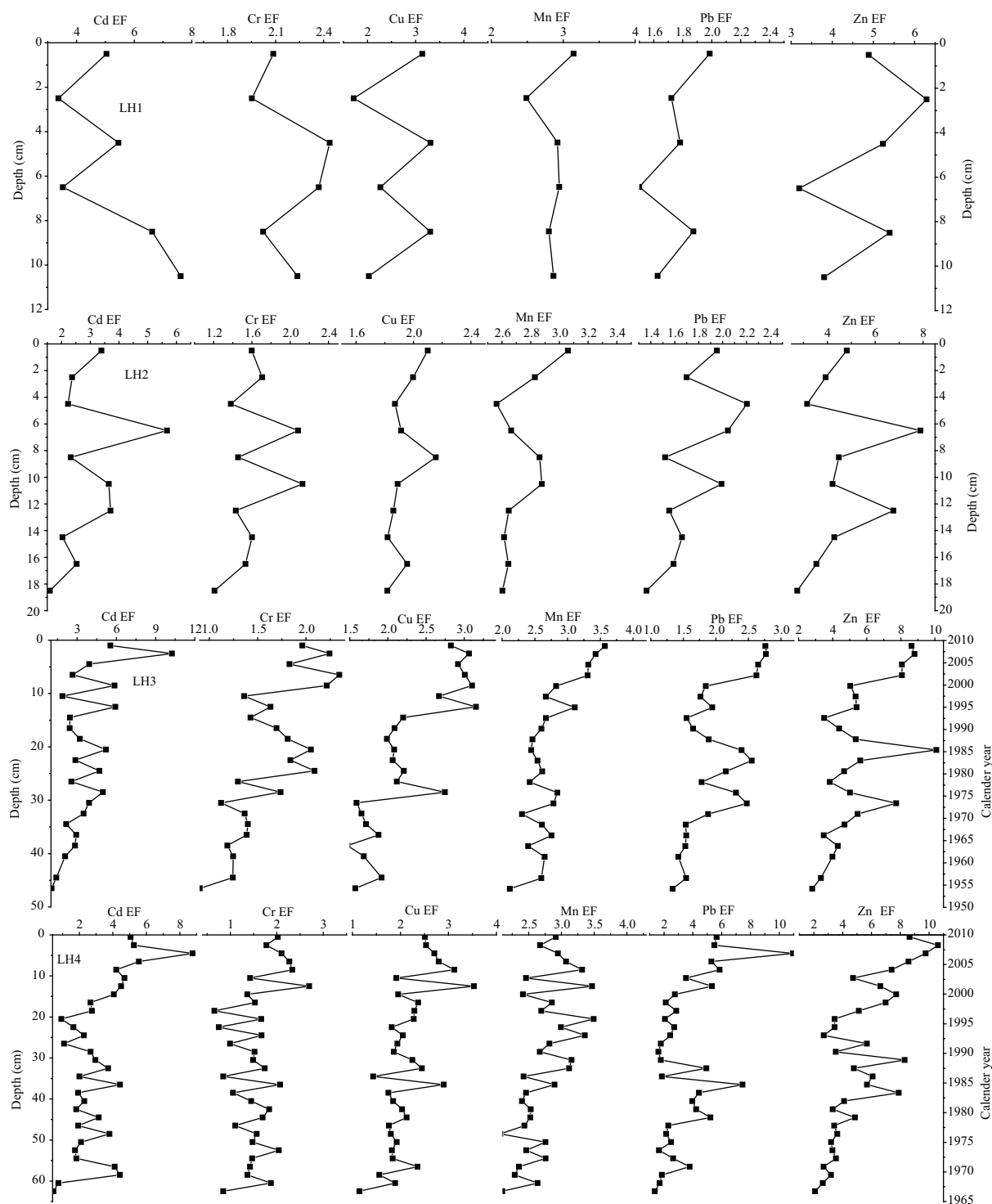
In general, EF values from 0.5 to 2.0 can be considered to be in the range of natural variability, indicating that the sediment is unaffected by human activities, and EF values greater than 2.0 represent heavy metal enrichment and contamination of the sediment as a result of significant anthropogenic input (Sutherland, 2000; Shumilin et al., 2002; Liaghati et al., 2003).

**Figure 5** summarizes the EFs calculated in this study, and trace metals are divided here in two categories. For the first group, the EFs of Cr and Cu range from 1.5 to 2.3, suggesting no or minimal anthropogenic enrichment. Both EFs present relatively low and constant values over the entire profiles. These results underline the facts that these elements have primarily lithogenic origins and that no significant anthropogenic source was detectable in all of the studied sites. For the other metals, the EF values range between 2 and 13 and follow the order Zn > Pb > Cd > Mn, indicating predominantly anthropogenic input of these elements. The enrichment in sediments of the tidal flat is most likely due to sewage disposal (Liu et al., 2003; Shi et al., 2010). Domestic waste could be a major source of pollution in the sediments. In addition, the emissions from the industrial activities in Lianyungang City significantly influenced the environment of the study area. The sediment cores are located close to the estuary through which the Linhong River flows (**Fig. 1**). This river runs through Lianyungang City, which produces large amounts of industrial and domestic waste that discharge

**Table 2** Pearson correlation coefficients between contents of heavy metals and mean grain size in tidal flat sediment from Haizhou Bay

	Mean grain size	Cd	Cr	Cu	Mn	Pb	Zn
Mean grain size	1						
Cd	0.15	1					
Cr	0.21	0.68*	1				
Cu	0.32	0.67*	0.75*	1			
Mn	0.222	0.56*	0.65*	0.81*	1		
Pb	-0.12	0.46*	0.36*	0.40*	0.34*	1	
Zn	-0.20	0.61*	0.41*	0.46*	0.42*	0.52*	1

\* Coefficients at 0.05 significance level (2-tailed).



**Fig. 5** Profiles of enrichment factors (EF) of heavy metals in sediment cores from tidal flat in Haizhou Bay.

into the river and cause serious heavy metal pollution in the river water, sediment and soil (Zhang et al., 2008). For example, there are many chemical and pharmaceutical factories near the Linhong River and the coastal zone of Haizhou Bay that heavily pollute the soil with heavy metals. In particular, anthropogenic Pb is considered to be primarily distributed by non-point sources such as

atmospheric transport and precipitation, which carry Pb that originated from coal burning in power generation plants and emission from ships and traffic (Li et al., 2001; Liu et al., 2003; Ip et al., 2004; Shi et al., 2010). Additionally, these enriched metals may have originated from non-point sources such as agricultural pollution (e.g., fertilizers and livestock manure) and atmospheric transport

and precipitation, which contain heavy metals as a result of coal burning in power generation plants, the use of leaded gasoline and other industrial activities (Tug and Duman, 2010). Moreover, domestic and industrial garbage has accumulated in the coastal zone; these deposits then act as point sources of metals (Pb, Zn and Cu) in soils. In addition, the burning of fuel results in widespread lead contamination (Santos et al., 2005)

## 2.5 Pollution assessment

The index of geoaccumulation ( $I_{geo}$ ) was calculated to determine the level of pollution using the equation (Muller, 1969):

$$I_{geo} = \log_2(C_n/1.5B_n) \quad (4)$$

where,  $C_n$  (mg/kg) is the measured concentration of the element  $n$ , and  $B_n$  (mg/kg) is the geochemical background value of element  $n$  in the coastal soil of Jiangsu (Xia et al., 1987). The factor 1.5 is introduced to include possible variations in the background values due to lithogenic effects. The following classification is given for  $I_{geo}$  (Muller, 1969):  $I_{geo} \leq 0$  essentially unpolluted,  $0 < I_{geo} < 1$  unpolluted to moderately polluted,  $1 < I_{geo} \leq 2$  moderately polluted,  $2 < I_{geo} \leq 3$  moderately to strongly polluted,  $3 < I_{geo} \leq 4$  strongly polluted,  $4 < I_{geo} \leq 5$  strongly to extremely polluted, and  $I_{geo} > 5$  extremely polluted. The  $I_{geo}$  values for Cr, Cu and Mn indicate that the study areas are practically unpolluted by these metals. The high values for Cd, Pb, and Zn in the tidal flat show that the area is unpolluted to moderately polluted by these metals (Fig. 6). The  $I_{geo}$  values for all of these metals are rated as not significantly harmful, but the tendency for pollution has already been established. Overall, the  $I_{geo}$  values reflect the general contamination tendency already showed by the EF values, and hence, the area can be considered to exhibit moderate contamination by Cd, Pb and Zn and weakly contaminated or uncontaminated with respect to Cr, Cu and Mn.

## 2.6 Modern industrial impact

Dated sediment cores were used to reconstruct the historical metal input to coastal sediments. In this study, we used heavy metal elements of anthropogenic origin to reconstruct the effects of modern industry on sediment quality. Figures 4 and 5 show the historic profiles of the concentrations and EF values of heavy metals in the cores. In these figures, the vertical axes of cores LH3 and LH4 were expressed as the estimated age of the sediment as determined by the  $^{210}\text{Pb}$  method. The surface sediment was from 2010, and the bottom layers were deposited in 1952 and 1967 for cores LH3 and LH4, respectively. The layer with the highest concentration of heavy metals, at a depth of 5–20 cm, was from the period 1990–2005.

In general, the EFs of heavy metals in the core samples tended to increase from the bottom upward (Fig. 5). This

trend points to the deposition of anthropogenic heavy metals related to the increase of agricultural activity, urbanization and industrialization near Haizhou Bay in the time period spanned by the sediment cores. The history of heavy metal pollution in the tidal flat of Haizhou Bay correlates with the economic growth of Lianyungang City. From 1949 to 1990, the gross industrial output value (GIOV) increased only from 50 million to 559 million RMB Yuan (Fig. 7) (SBJ, 2010). This period of slow economic development in Lianyungang is reflected by the slowly rising metal contents in the core samples; in 1978, the “Reform and Open” policy was initiated, which resulted in the higher levels of heavy metals in the 1990s.

The increasing trend of heavy metal pollution continued with the rapid economic growth in the 1990s. The peak heavy metal contamination occurred from the 1990s to the 2000s, and this time period corresponds to the increase in gross domestic product (GDP) and in heavy metal waste output of Lianyungang City (Fig. 6). Since the 1990s, Lianyungang has experienced its most rapid industrialization (urbanization). In this period, the GDP has dramatically increased from 4930 million to 45,600 million RMB Yuan (Fig. 7) (SBJ, 2010), and at the same time, the demand for energy from fossil fuels for power generation, car and shipping transportation, as well as the corrosion of building materials, urban traffic-related emissions and industrial and domestic waste, have drastically increased the contaminant level and further compromised the environment.

As a consequence of legal restrictions and improvements in waste processing technology, the concentration of heavy metals in the sediment is assumed to have decreased. However, as shown in Fig. 4 and Table 2, the current heavy metal pollution in the tidal flat of Haizhou Bay is still higher than in other sediments, such as local and global background sediments, marine sediments in Hong Kong and sediments in the Yellow Sea, and is higher than the maximum levels in sediments of the pre-industrial period (Xia et al., 1987; Taylor, 1964; EPD, 1992; Chi and Yan, 2007). These trends indicate that the contaminated environment could continue to be a hazard to the health of wildlife dwelling in the tidal flat of Haizhou Bay.

It is not sufficient to reconstruct the heavy metal enrichment and pollution history based on only a few sediment cores and limited dating in such a complicated depositional environment as an estuarine and coastal zone. However, this study clearly demonstrates that heavy metal concentrations have increased over the last several decades and have resulted in significant heavy metal enrichment and pollution in the sediments of the tidal flat of Haizhou Bay in recent years.

## 3 Conclusions

The short core sediments from the intertidal flat of Haizhou Bay consisted of fine-grained materials. The high mean

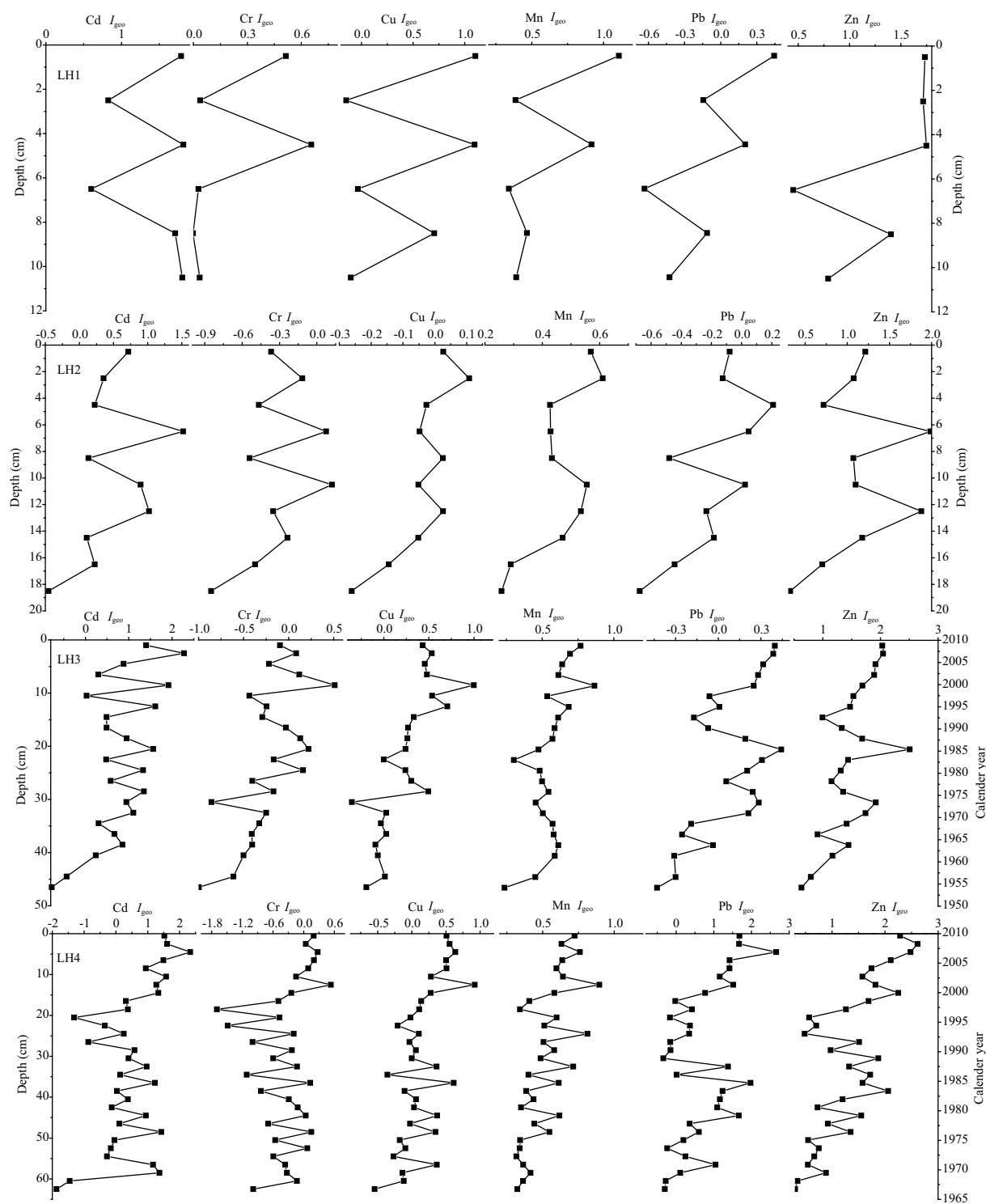
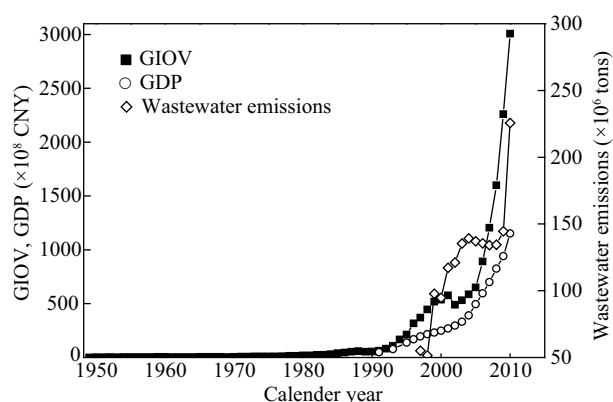


Fig. 6 Index of geoaccumulation for heavy metals in sediment cores.

sedimentation rates of 0.85 and 1.5 cm/yr were calculated for cores LH3 and LH4, respectively, and with the aid of with  $^{210}\text{Pb}$  dating, these sediments were used to reveal the recent evolution of heavy metal pollution. This study of trace metal abundances and distributions in the sediment cores shows that the historic (< 50 years) metal pollution,

which originated from anthropogenic activities rather than lithogenic geochemical sources, can be directly related to the socio-economic development of the area.

This study has confirmed that the heavy metal pollution in Haizhou Bay has been and continues to be strongly influenced by human activity in the coastal area.



**Fig. 7** Historical economic development and wastewater emissions of Lianyungang in past 50 years. GIOV and GDP represent gross industrial output value and gross domestic product, respectively.

### Acknowledgments

This work was supported by the Open Fund of State Key Laboratory of Pollution Control and Resources Reuse (No. PCRRF11024), the Fostering Project of Major Research Plan of National Natural Sciences Foundation of China (No. 91025022), the Natural Science Foundation of Huaihai Institute of Technology (No. Z2011001), the Priming Scientific Research Foundation for the Junior Teachers in Huaihai Institute of Technology (No. KQ09041), the Project of Innovation from Undergraduate in Jiangsu Province (No. SY 201311641107001), and the project funded by the Priority Academic Program Development of Jiangsu Higher Education Institutions.

### References

- Andersen T J, Mikkelsen O A, Moller A L, Morten P, 2000. Deposition and mixing depths on some European intertidal mudflats based on <sup>210</sup>Pb and <sup>137</sup>Cs activities. *Continental Shelf Research*, 20(12): 1569–1591.
- Aksu A, Balks N, Ersan M S, Muftuoglu A E, Apak R, 2010. Biogeochemical cycle of arsenic and calculating the enrichment factor by using Li element. *Environmental Geochemistry and Health*, 32(4): 303–306.
- Bellucci L G, Frignani M, Paolucci M, Ravanelli M, 2002. Distribution of heavy metals in sediments of the Venice Lagoon: the role of the industrial area. *Science of the Total Environment*, 295(1-3): 35–49.
- Benoit G, Wang E X, Nieder W C, Levandowsky M, Breslin V T, 1999. Sources and deposition in history of heavy metal contamination and sediment Tivoli South Bay, Hudson River, New York. *Estuaries*, 22(2): 167–178.
- Cantwell M G, King J W, Burgess R M, Appleby P G, 2007. Reconstruction of contaminant trends in a salt wedge estuary with sediment cores dated using a multiple proxy approach. *Marine Environmental Research*, 64(2): 225–246.
- Chen B L, 2006. Environmental evolution and countermeasure research of Lianyungang offshore area. D. Sc. Thesis, East China Normal University, China. 53–103.
- Chi Q H, Yan M C, 2007. Handbook of Elemental Abundance for Applied Geochemistry. The Geological Publishing House,

- Beijing. 96–97.
- Cobelo-Garcia A, Prego R, 2003. Heavy metal sedimentary record in a Galician Ria (NW Spain): background values and recent contamination. *Marine Pollution Bulletin*, 46(10): 1253–1262.
- Eisenrhcil S J, Metzger N A, Urman N R, 1986. Response of atmospheric lead to decreased use of lead in gasoline. *Environmental Science and Technology*, 20(2): 171–174.
- EPD (Environmental Protection Department), 1992. Marine Water Quality in Hong Kong. Environmental Protection Department of Hong Kong, Hong Kong.
- Feng H, Han X, Zhang W, Yu L, 2004. A preliminary study of heavy metal contamination in Yangtze River intertidal zone due to urbanization. *Marine Pollution Bulletin*, 49 (11–12): 910–915.
- Flynn W W, 1968. The determination of low level of polonium-210 in environmental materials. *Analytica Chimica Acta*, 43(1): 221–227.
- Hakanson L, 1980. An ecological risk index for aquatic pollution control: a sedimentological approach. *Water Research*, 14(1): 975–1000.
- Harbison P, 1986. Mangrove muds: a sink and a source for trace metals. *Marine Pollution Bulletin*, 17(6): 246–250.
- Hartmann P C, Quinn J G, Cairns R W, King J W, 2005. Depositional history of organic contaminants in Narragansett Bay, Rhode Island, USA. *Marine Pollution Bulletin*, 50(4): 388–395.
- Ip C C M, Li X D, Zhang G, Farmer J G, Wai O W H, Li Y S, 2004. Over one hundred years of trace metal fluxes in the sediments of the Pearl River Estuary, South China. *Environmental Pollution*, 132(1): 157–172.
- Lacerda L D, Martinelli L A, Rezende C A, Mozetto A A, Ovalle A R C, Victoria R L et al., 1988. The fate of heavy metals in suspended matter in a mangrove creek during a tidal cycle. *Science of the Total Environment*, 75(2-3): 169–180.
- Laluraj C M, Nair S M, 2006. Geochemical index of trace metals in the surficial sediments from the western continental shelf of India, Arabian Sea. *Environmental Geochemistry and Health*, 28(6): 509–518.
- Lee S V, Cundy A B, 2001. Heavy metal contamination and mixing processes in sediments from the Humber Estuary, Eastern England. *Estuarine Coastal and Shelf Science*, 53(5): 619–636.
- Li Q S, Wu Z F, Chu B, Zhang N, Cai S S, Fang J H, 2007. Heavy metals in coastal wetland sediments of the Pearl River Estuary, China. *Environmental Pollution*, 149(2): 158–164.
- Li Y, Feng Z H, Li G Q, Yan B L, 2010. The estimation of source of heavy metal contamination and assessment in marine sediments in Lianyungang area. *Oceanologica Et Limnologia Sinica*, 41(1): 829–833.
- Liaghati T, Preda M, Cox M, 2003. Heavy metal distribution and controlling factors within coastal plain sediments, Bells Creek catchment, southeast Queensland, Australia. *Environmental International*, 29(7): 935–948.
- Liu W X, Li X D, Shen Z G, Wang D C, Wai O W H, Li Y S, 2003. Multivariate statistical study of heavy metal enrichment in sediments of the Pearl River Estuary. *Environmental Pollution*, 121(3): 377–388.
- Li X D, Wai O W H, Li Y S, Coles B J, Ramsey M H, Thornton I, 2000. Heavy metal distribution in sediment profiles of the

- Pearl River estuary, South China. *Applied Geochemistry*, 15(5): 567–581.
- Li X D, Shen Z G, Wai O W, Li Y S, 2001. Chemical forms of Pb, Zn and Cu in the sediment profiles of the Pearl River Estuary. *Marine Pollution Bulletin*, 42(2): 215–223.
- Lu X, 2007. A note on removal of the compaction effect for the  $^{210}\text{Pb}$  method. *Applied Radiation and Isotopes*, 65(1): 142–146.
- Maanan M, 2008. Heavy metal concentrations in marine mollusks from the Moroccan coastal region. *Environmental Pollution*, 153(1): 176–183.
- Mil-Homens M, Stevens R L, Abrantes F, Cato I, 2006. Heavy metal assessment for surface sediments from three areas of the Portuguese continental shelf. *Continental Shelf Research*, 26(10): 1184–1205.
- Muller G, 1969. Index of geoaccumulation in sediments of the Rhine river. *Geojournal*, 2(3): 108–110.
- Munksgaard N C, Lim K, Parry D L, 2003. Rare earth elements as provenance indicators in North Australian estuarine and coastal marine sediments. *Estuarine, Coastal and Shelf Science*, 57(3): 399–409.
- Natesan U, Seshan B R, 2010. Vertical profile of heavy metal concentration in core sediments of Buckingham canal, Ennore. *Indian Journal of Geo-Marine Sciences*, 40(1): 83–97.
- Niu H Y, Deng W J, Wu Q H, Chen X G, 2009. Potential toxic risk of heavy metals from sediment of the Pearl River in South China. *Journal of Environmental Sciences*, 21(8): 1053–1058.
- Peng J, Chen S L, 2010. Analysis on sedimentary characteristics and environments in nearshore of Lianyungang. *Advances in Marine Science*, 28(2): 445–454.
- Rubio B, Nombela M A, Vilas F, 2000. Geochemistry of major and trace elements in sediments of the Ria de Vigo (NW Spain): an assessment of metal pollution. *Marine Pollution Bulletin*, 40(11): 968–980.
- Sanders C J, Smoak J M, Naidu A S, Sanders L M, Patchineelam S R, 2010. Organic carbon burial in a mangrove forest, margin and intertidal mud flat. *Estuarine Coastal and Shelf Science*, 90(2): 168–172.
- Santos I R, Silva-Filho E V, Schaefer C E, Albuquerque-Filho M R, Campos L S, 2005. Heavy metal contamination in coastal sediments and soils near the Brazilian Antarctic Station, King George Island. *Marine Pollution Bulletin*, 50(2): 185–194.
- SBJ (Statistical Bureau of Jiangsu), 2010. Statistical Yearbook of Jiangsu. Statistics Bureau of Jiangsu, Nanjing, China. 10–20.
- Schiff K C, Weisberg S B, 1999. Iron as a reference element for determining trace metal enrichment in Southern California coastal shelf sediments. *Marine Environmental Research*, 48(2): 161–176.
- Selvaraj K, Mohan V R, Szefer P, 2004. Evaluation of metal contamination in coastal sediments of the Bay of Bengal, India: geochemical and statistical approaches. *Marine Pollution Bulletin*, 49(3): 174–185.
- Shi Q, Thomas L, Peter R, Zhou D, Jan H, 2010. Geochemical sources, deposition and enrichment of heavy metals in short sediment cores from the Pearl River Estuary, Southern China. *Journal of Marine Systems*, 82(Suppl.): 1–15.
- Shumilin E N, Carriquiry J D, Camacho-Ibar V F, Sapozhnikov D, Kalmykov S, Sanchez A et al., 2002. Spatial and vertical distributions of elements in sediments of the Colorado River delta and Upper Gulf of California. *Marine Chemistry*, 79(3-4): 113–131.
- Spencer K L, Cundy A B, Croudace I W, 2003. Heavy metal distribution and early-diagenesis in salt marsh sediments from the Medway Estuary, Kent, UK. *Estuarine, Coastal and Shelf Science*, 57(1-2): 43–54.
- Sutherland R A, 2000. Bed sediment-associated trace metals in an urban stream, Oahu, Hawaii. *Environmental Geology*, 39(6): 611–627.
- Tang C W, Ip C C, Zhang G, Shin P K S, Qian P Y, Li X D, 2008. The spatial and temporal distribution of heavy metals in sediments of Victoria Harbour, Hong Kong. *Marine Pollution Bulletin*, 57(1): 816–825.
- Taylor S R, 1964. Abundance of chemical elements in the continental crust: a new table. *Geochimica et Cosmochimica Acta*, 28(8): 1273–1285.
- Tug G N, Duman F, 2010. Heavy metal accumulation in soils around a salt lake in Turkey. *Pakistan Journal of Botany*, 42(4): 2327–2333.
- Tanner P A, Leong L S, Pan S M, 2000. Contamination of heavy metals in marine sediment cores from Victoria Harbour, Hong Kong. *Marine Pollution Bulletin*, 40(9): 769–779.
- Xia Z L, Li S Z, Li T F, 1987. Soil Background Value and its Researching Methods. Meteorology Press, Beijing. 314–316.
- Zhang C Y, Feng X L, Chen B L, 2008. Analysis of heavy metal pollution in cores from the south Haizhou Bay. *Marine Geology & Quaternary Geology*, 5(2): 37–43.
- Zhang J, Liu C L, 2002. Riverine composition and estuarine geochemistry of particulate metals in China-weathering features, anthropogenic impact and chemical fluxes. *Estuarine, Coastal and Shelf Science*, 54(6): 1051–1070.
- Zourarah B, Maanan M, Carruesco C, Aajjane A, Mehdi K, Conceicao Freitas M, 2007. Fifty-year sedimentary record of heavy metal pollution in the lagoon of Oualidia (Moroccan Atlantic coast). *Estuarine, Coastal and Shelf Science*, 72 (1-2): 359–369.

# JOURNAL OF ENVIRONMENTAL SCIENCES

环境科学学报(英文版)  
(<http://www.jesc.ac.cn>)

## Aims and scope

*Journal of Environmental Sciences* is an international academic journal supervised by Research Center for Eco-Environmental Sciences, Chinese Academy of Sciences. The journal publishes original, peer-reviewed innovative research and valuable findings in environmental sciences. The types of articles published are research article, critical review, rapid communications, and special issues.

The scope of the journal embraces the treatment processes for natural groundwater, municipal, agricultural and industrial water and wastewaters; physical and chemical methods for limitation of pollutants emission into the atmospheric environment; chemical and biological and phytoremediation of contaminated soil; fate and transport of pollutants in environments; toxicological effects of terrorist chemical release on the natural environment and human health; development of environmental catalysts and materials.

## For subscription to electronic edition

Elsevier is responsible for subscription of the journal. Please subscribe to the journal via <http://www.elsevier.com/locate/jes>.

## For subscription to print edition

China: Please contact the customer service, Science Press, 16 Donghuangchenggen North Street, Beijing 100717, China. Tel: +86-10-64017032; E-mail: [journal@mail.sciencep.com](mailto:journal@mail.sciencep.com), or the local post office throughout China (domestic postcode: 2-580).

Outside China: Please order the journal from the Elsevier Customer Service Department at the Regional Sales Office nearest you.

## Submission declaration

Submission of an article implies that the work described has not been published previously (except in the form of an abstract or as part of a published lecture or academic thesis), that it is not under consideration for publication elsewhere. The submission should be approved by all authors and tacitly or explicitly by the responsible authorities where the work was carried out. If the manuscript accepted, it will not be published elsewhere in the same form, in English or in any other language, including electronically without the written consent of the copyright-holder.

## Submission declaration

Submission of the work described has not been published previously (except in the form of an abstract or as part of a published lecture or academic thesis), that it is not under consideration for publication elsewhere. The publication should be approved by all authors and tacitly or explicitly by the responsible authorities where the work was carried out. If the manuscript accepted, it will not be published elsewhere in the same form, in English or in any other language, including electronically without the written consent of the copyright-holder.

## Editorial

Authors should submit manuscript online at <http://www.jesc.ac.cn>. In case of queries, please contact editorial office, Tel: +86-10-62920553, E-mail: [jesc@263.net](mailto:jesc@263.net), [jesc@rcees.ac.cn](mailto:jesc@rcees.ac.cn). Instruction to authors is available at <http://www.jesc.ac.cn>.

## Journal of Environmental Sciences (Established in 1989)

Vol. 25 No. 7 2013

<b>Supervised by</b>	Chinese Academy of Sciences	<b>Published by</b>	Science Press, Beijing, China
<b>Sponsored by</b>	Research Center for Eco-Environmental Sciences, Chinese Academy of Sciences	<b>Distributed by</b>	Elsevier Limited, The Netherlands
<b>Edited by</b>	Editorial Office of Journal of Environmental Sciences P. O. Box 2871, Beijing 100085, China Tel: 86-10-62920553; <a href="http://www.jesc.ac.cn">http://www.jesc.ac.cn</a> E-mail: <a href="mailto:jesc@263.net">jesc@263.net</a> , <a href="mailto:jesc@rcees.ac.cn">jesc@rcees.ac.cn</a>	<b>Domestic</b>	Science Press, 16 Donghuangchenggen North Street, Beijing 100717, China Local Post Offices through China
<b>Editor-in-chief</b>	Hongxiao Tang	<b>Foreign</b>	Elsevier Limited <a href="http://www.elsevier.com/locate/jes">http://www.elsevier.com/locate/jes</a>
<b>CN 11-2629/X</b>	<b>Domestic postcode: 2-580</b>	<b>Printed by</b>	Beijing Beilin Printing House, 100083, China
		<b>Domestic price per issue</b>	<b>RMB ¥ 110.00</b>

ISSN 1001-0742

