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## Review

# Removal of antibiotic-resistant genes during drinking water treatment: A review

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### ABSTRACT

Once contaminate the drinking water source, antibiotic resistance genes (ARGs) will propagate in drinking water systems and pose a serious risk to human health. Therefore, the drinking water treatment processes (DWTPs) are critical to manage the risks posed by ARGs. This study summarizes the prevalence of ARGs in raw water sources and treated drinking water worldwide. In addition, the removal efficiency of ARGs and related mechanisms by different DWTPs are reviewed. Abiotic and biotic factors that affect ARGs elimination are also discussed. The data on presence of ARGs in drinking water help come to the conclusion that ARGs pollution is prevalent and deserves a high priority. Generally, DWTPs indeed achieve ARGs removal, but some biological treatment processes such as biological activated carbon filtration may promote antibiotic resistance due to the enrichment of ARGs in the biofilm. The finding that disinfection and membrane filtration are superior to other DWTPs adds weight to the advice that DWTPs should adopt multiple disinfection barriers, as well as keep sufficient chlorine residuals to inhibit re-growth of ARGs during subsequent distribution. Mechanistically, DWTPs obtain direct and indirect ARGs reduction through DNA damage and interception of host bacteria of ARGs. Thus, escaping of intracellular ARGs to extracellular environment, induced by DWTPs, should be avoided. This review provides the theoretical support for developing efficient reduction technologies of ARGs. Future study should focus on ARGs controlling in terms of transmissibility or persistence through DWTPs due to their biological related nature and ubiquitous presence of biofilm in the treatment unit.

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## Introduction

Antibiotic disposal into the environment enhances selective pressures that lead to the emergence, proliferation, and prevalence of antibiotic resistant bacteria (ARB) and antibiotic resistance genes (ARGs) (Luo and Zhou, 2008; Su et al., 2014).

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Most of the antibiotics currently in use can acquire their corresponding ARB from the environment (Xu et al., 2011). Furthermore, ARGs can be transmitted within and between species through vertical gene transfer (VGT) and horizontal gene transfer (HGT) (Dodd, 2012; Li et al., 2019; Lu et al., 2020). After bacterial death, intracellular ARGs are released into the environment where they exist for extended periods of time. Notably, ARGs can be transferred to other non-bacterial organisms, including human beings, to inhibit their susceptibilities to antibiotics (Li et al., 2018b). Due to the antibiotic resistance imposed by ARGs and their threat to environmental safety and human health, they have been recognized as emerging environmental contaminants (Pruden et al., 2006).

Various ARGs at different concentration have been detected in air (Gibbs et al., 2004; Xie et al., 2019; Zhang et al., 2019b), soil (Lu and Lu, 2019; Zhao et al., 2017), medical wastewater (Szekeress et al., 2017), aquaculture wastewater (Chen et al., 2017), surface water (Dang et al., 2017; Stange et al., 2019b; Yang et al., 2018), and drinking water (Fernando et al., 2016; Gomes Freitas et al., 2017). Drinking water is an important potential route of ARGs exposure to humans. Therefore, limiting ARGs and their spread in drinking water systems is vital for human health. It is difficult and costly to decrease the concentration levels of ARGs in source waters and to renovate water distribution networks for drinking water systems. A reliable way of controlling ARGs in drinking water is through the interception or stabilization function of waterworks. Therefore, it is necessary to summarize how and why the currently applied drinking water treatment technologies remove ARGs.

This work reviews the presence of ARGs in drinking water, removal efficiency and mechanism of ARGs by different treatment processes in waterworks, and influencing factors on ARGs elimination based on the latest literature. The route discussed in this paper begins from the abundance of ARGs in raw source water, finished water of waterworks, and tap water, to draw a general picture of how waterworks in ARGs removal. Then the removal efficiency of ARGs for specific treatment process (coagulation/sedimentation, clarification, ozonation, biological activated carbon filtration, membrane filtration, and disinfection) is discussed, followed by the analysis of related mechanism. Finally, special attention is given to the abiotic and abiotic factors influencing the reduction of ARGs from the perspective of how waterworks minimize the induction of ARGs generation and promotion of ARGs propagation brought by drinking water treatment processes themselves.

## 1. ARGs in raw water sources and treated drinking water

Antibiotic resistance genes in drinking water have been attributed to raw water sources. Six ARGs were detected in raw water sources of Michigan and Ohio in America (Xi et al., 2009). In another study which conducted in Louisiana, the raw water sources of were found to contain *sul1* and *tetA* genes (Scott et al., 2015). Studies conducted in other countries including Nigeria (Adesoji et al., 2017), Canada (Fernando et al., 2016), Korea (Son et al., 2018), Japan (Nguyen et al., 2019), Tanzania (Iyimo et al., 2016),

Poland (Koniuszewska et al., 2020), Germany (Stoll et al., 2012; Voigt et al., 2020), Australia (Stoll et al., 2012), UK (Dhanji et al., 2011), India (Ahammad et al., 2014), France (Laroche et al., 2009), and Russia (Sazykin et al., 2019) also discovered the presence of ARGs in raw water sources. In China, aminoglycoside,  $\beta$ -lactam, sulfonamide, FCA (fluroquinolone, quinolone, florfenicol, chloramphenicol, and amphenicol), macrolide-lincosamide-streptogramin B (MLSB), other/efflux, vancomycin and tetracycline resistance genes were found in the Qiantang river (Xu et al., 2016). Seventeen ARGs were detected in the Yellow river of China (Lu et al., 2018). The economically developed regions of China exhibited worse outcomes regarding ARGs in drinking water source. For example, 27 ARGs were detected in raw water in Pearl River Delta (Su et al., 2018). The presence of ARGs in raw water sources worldwide is summarized in Table 1.

Naturally, the ARG pollution of drinking water source will spread to the treated drinking water. Numerous studies have revealed that the ARGs were detected in treated drinking water including finished water of waterworks, water in drinking water distribution system, tap water, cistern water of residential building, and spring (Table 1). The bad news is the frequent detection of ARGs in tap water which directly contacts with consumers. In Poland, Canada, and California, *bla*, *sul*, *tet*, *amp* and *erm* resistance genes were detected in tap water. In Singapore and China, 265 ARGs belonging to 17 different types were found in tap water obtained from several cities (Ma et al., 2019). Among these ARGs, multidrug, bacitracin, and aminoglycoside resistance genes were the most dominant. Twenty-four ARGs were also found in tap water collected from 71 cities of 31 provincial level administrative regions of China (Zhang et al., 2020). Among the ARGs, those conferring resistance to sulfonamides exhibited the highest distribution and abundance followed by those conferring resistance to aminoglycosides and tetracyclines. The fact that ARGs are frequently detected in raw water sources and treated drinking water after waterworks indicates that ARGs pollution in drinking water is universal and deserves a high priority.

## 2. The efficiency of ARGs removal during the drinking water treatment processes

Drinking water treatment processes include pretreatment processes (pre-oxidation, biological pretreatment, adsorption pretreatment with powder activated carbon, etc.), conventional treatment processes, and advanced treatment processes. When the removal targets are suspended matters, colloidal substances, bacterias, and viruses, conventional treatment processes (i.e., coagulation/sedimentation, clarification, sand filtration, and disinfection) are suggested. If there are micropollutants such as ammonia, pesticides, and heavy metals in the raw water, conventional treatment processes are recommended to run with assistance of advanced treatment processes (granular activated carbon adsorption, granular activated carbon adsorption combined with sand filtration, BAC, O<sub>3</sub>-BAC, membrane filtration etc.). Table 2 lists mainstream drinking water treatment processes in different countries. Up to now, conventional treatment processes still play a major role in the production of drinking water around the world, and

**Table 1 – Presence of ARGs in raw source water and drinking water worldwide.**

Source	Region	ARGs Type	Abundance (copies/L)	Refs.
Raw source water	Louisiana (America)	sul1, tetA	Unknown	Scott et al. (2015)
	Michigan and Ohio (America)	bla <sub>TEM</sub> , bla <sub>SHV</sub> , sul1, sul2, cat, cmr	10 <sup>2</sup> –10 <sup>6</sup>	Xi et al. (2009)
	America	sul1, sul2, intI1	10 <sup>5</sup> –10 <sup>2</sup> (Relative abundance)	Rocha et al. (2019)
	Lake Michigan (America)	Sul1, sul2, mexB, acrA, acrB, qacH, intI1	10 <sup>6</sup> –10 <sup>8</sup>	Kappell et al. (2019)
	Southwestern Nigeria	sul1, sul2	Unknown	Adesoji et al. (2017)
	Island Lake Region of Manitoba (Canada)	ampC, tetA, mecA	1–7 (copy No./ng DNA)	Fernando et al. (2016)
	Cheongju and Cheonan (Korea)	tetA, tetX, sul1, ermB, ermC, qnrS, aac(6')-lb-cr, floR, bla <sub>TEM</sub> , bla <sub>SHV</sub> , oqxA, intI1	10 <sup>3</sup> –10 <sup>11</sup>	Son et al. (2018)
	Tama River and Lake Kasumigaura (Japan)	bla <sub>TEM</sub> , sul1, tetA, ere(A), intI1	10 <sup>4</sup> –10 <sup>9</sup>	Nguyen et al. (2019)
	Northern Tanzania	bla <sub>TEM-1</sub> , bla <sub>CTX-M</sub> , bla <sub>SHV-1</sub> , tetA, tetB	Unknown	Lyimo et al. (2016)
	River Thames (London, UK)	bla <sub>CTX-M-14</sub>	Unknown	Dhanji et al. (2011)
	Pilica river (Poland)	β-lactam, tetracycline, MLS, fluoroquinolones and sulfonamide resistance genes	Unknown	Koniuszewska et al. (2011); (Koniuszewska et al., 2020)
	River Brisbane, Rhine and Danube (Australia and Germany)	Sulfonamide, trimethoprim, β-lactam, aminoglycosides, chloramphenicol, tetracycline, macrolide, glycopeptide resistance genes	Unknown	Stoll et al. (2012)
	Germany	bla <sub>OXA-58,mcr</sub>	Unknown	Voigt et al. (2020)
	Ganges river (India)	bla <sub>NDM-1</sub> , bla <sub>OXA</sub> , tetM, tetW, tetQ	10 <sup>3</sup> –10 <sup>9</sup>	Ahammad et al. (2014)
Finished water of DWTP	Seine (France)	intI1 and intI2	Unknown	Laroche et al. (2009)
	Don River (Russia)	OXA-48, VanB and CTX-M	Unknown	(Sazykin et al., 2019)
	Qiantang river (Hangzhou, China)	Aminoglycosides, β-lactams, sulfonamides, FCA, MLSB, other/efflux, vancomycin and tetracycline resistance genes	10 <sup>7</sup> –10 <sup>10</sup>	Xu et al. (2016)
	Pearl River (Guangzhou, China)	Sulfonamides, tetracycline, chloramphenicol, quinolones and macrolides resistance genes	10 <sup>4</sup> –10 <sup>11</sup>	Su et al. (2018)
	Tai lake (China)	sul1, sul2, tetC, tetG, tetX, tetA, tetB, tetO, tetM, tetW	10 <sup>3</sup> –10 <sup>9</sup>	Guo et al. (2014)
	Yellow River (Lanzhou, Yinchuan, Hohhot, Zhengzhou, Jinan, Dongying, China)	Aminoglycosides, macrolide, tetracycline, sulfonamides resistance genes	<10 <sup>8</sup>	Lu et al. (2018)
	Qiantang river (Hangzhou, China)	Aminoglycosides, β-lactams, sulfonamides, FCA, MLSB, other/efflux, vancomycin and tetracycline resistance genes	10 <sup>5</sup> –10 <sup>8</sup>	Xu et al. (2016)
	Pearl River (Guangzhou, China)	Sulfonamides, tetracycline, chloramphenicol, quinolones and macrolides resistance genes	10 <sup>4</sup> –10 <sup>10</sup>	Su et al. (2018)
	Tai Lake (China)	sul1, sul2, tetC, tetG, tetX, tetA, tetB, tetO, tetM, tetW	<10 <sup>8</sup>	Guo et al. (2014)
Wroclaw (Poland)	Yellow River (Lanzhou, Yinchuan, Hohhot, Zhengzhou, Jinan, Dongying, China)	Aminoglycosides, macrolide, tetracycline, sulfonamides resistance genes	<10 <sup>7</sup>	Lu et al. (2018)
	Wroclaw (Poland)	bla <sub>NDM</sub> , qnrS, qacH, intI1, bla <sub>TEM</sub> , ampC, tolA	Unknown	Siedlecka et al. (2020)

(continued on next page)

**Table 1 (continued)**

Source	Region	ARGs Type	Abundance (copies/L)	Refs.
Tap water	Michigan and Ohio (America)	<i>blaTEM</i> , <i>blaSHV</i> , <i>sul1</i> , <i>sul2</i> , <i>cat</i> , <i>cmr</i>	$10^3$ – $10^9$	<a href="#">Xi et al. (2009)</a>
	America	<i>tetA</i> , <i>sul1</i>	$10^{-4}$ – $10^{-2}$ (Relative abundance)	<a href="#">Rocha et al. (2019)</a>
	Hangzhou (China)	Aminoglycosides, $\beta$ -lactams, sulfonamides, FCA, MLSB, other/efflux, vancomycin and tetracycline resistance genes	< $10^8$	<a href="#">Xu et al. (2016)</a>
	Guangzhou (China)	Sulfonamides, tetracycline, chloramphenicol, quinolones and macrolides resistance genes	$10^4$ – $10^8$	<a href="#">Su et al. (2018)</a>
	71 cities (China)	Integration, quinolones, tetracycline, aminoglycosides, $\beta$ -lactam and macrolides resistance genes	< $10^7$ (Total ARGs)	<a href="#">Zhang et al. (2020)</a>
	Singapore and China (Hong Kong, Nanning, Jinzhong, Qingdao, Wugang, Shanghai, Linyi, Botou, Xigaze, Fujin, Yinchuan)	Multidrug, bacitracins, MLS, $\beta$ -lactam, aminoglycosides, sulfonamide, fosmidomycin, tetracycline, rifamycin, vancomycin, polymixin, trimethoprim, kasugamycin, quinolone, chloramphenicol, fosfomycin, puromycin resistance genes	$4 \times 10^{-2}$ – $10^0$ (Copies/cell)	<a href="#">Ma et al. (2019)</a>
	Wrocław (Poland)	<i>blaTEM</i> , <i>blaNDM</i> , <i>qnrB</i> , <i>qnrC</i> , <i>tetA</i> , <i>sul1</i> , <i>sul2</i> , <i>ermB</i> , <i>intI1</i>	Unknown	<a href="#">Siedlecka et al. (2020)</a>
	Island Lake Region of Manitoba (Canada)	<i>ampC</i> , <i>tetA</i>	1–2.5 (opy No./ng DNA)	<a href="#">Fernando et al. (2016)</a>
	California (Los Angeles, San Diego, Bakersfield, Fresno)	<i>blaSHV</i> , <i>sul1</i>	< $10^7$	<a href="#">Echeverria-Palencia et al. (2017)</a>
	Scotland	<i>sul1</i> and <i>sul2</i>	Unknown	<a href="#">Khan et al. (2016)</a>
Cistern water of residential building Spring	Northern Tanzania	<i>blaTEM_1</i> , <i>blaCTX-M</i> , <i>blaSHV-1</i> , <i>tetA</i> , <i>tetB</i>	Unknown	<a href="#">Lyimo et al. (2016)</a>
	Cajamarca (Peru)	<i>blaCTX-M-3</i>	Unknown	<a href="#">Larson et al. (2019)</a>
	Dhaka (Bangladesh)	<i>blaCTX-M-15</i> , <i>blaCMY-2</i> , <i>qnrS</i> , <i>qnrB</i>	Unknown	<a href="#">Talukdar et al. (2013)</a>
	Manitoba (Canada)	<i>blaSHV</i> , <i>blaTEM</i> , <i>blaCTX-M</i> , <i>blaOXA-1</i> , <i>blaCYM-2</i> , <i>blaKPC</i> , <i>blaOXA-48</i> , <i>blaNDM</i>	Unknown	<a href="#">Mi et al. (2019)</a>
	Portugal	<i>blaL1</i>	Unknown	<a href="#">Henriques et al. (2012)</a>

the most popular advanced treatment processes are O<sub>3</sub>-BAC and membrane filtration.

## 2.1. Pretreatment processes

The pretreatment processes are aimed at changing the nature and improving the treatability of pollutants by physical, chemical, and biological methods so that they can be effectively removed in the subsequent treatment process. Common raw water pretreatment processes include chemical oxidation, biodegradation, pH adjustment, precipitation among others. The existing reports regarding the removal of ARGs during the pretreatment process involve the pre-oxidation process that occurs before the coagulation/ precipitation or clarification processes. Therefore, this study laid much emphasis on the removal efficiency of ARGs during the pre-oxidation process.

### 2.1.1. Pre-chlorination

Pre-chlorination is the oldest and most common pre-oxidation method in waterworks. This technology utilizes active chlorine (hypochlorous acid, chlorine dioxide, chloramine) to oxidize and enhance the removal of pollutants (algae, iron, manganese, colloidal particles, etc.). In a previous study, it was found that after pre-chlorination in waterworks W, the absolute abundance of *sul1*, *sul2*, *tetG* and *tetA* genes were reduced by more than 1.7 log, while *tetO*, *tetM* and *tetW* genes were difficult to be eliminated even though their concentrations were low ([Guo et al., 2014](#)). However, the relative abundance of sulfonamide and tetracycline ARGs increased at least 1 log after pre-chlorination in waterworks W.

### 2.1.2. Pre-ozonation

The ozonation technology has been applied in water treatment procedures since the 1970s. This process enhances the

**Table 2 – Mainstream drinking water treatment processes in different countries.**

Countries	Common drinking water treatment processes	Refs.
UK	Surface water → Pre-ozonation → Coagulation → Sedimentation → Sand filtration → O <sub>3</sub> /BAC → Chlorine disinfection	Xu et al. (2004); ( <a href="#">Xu, 2004</a> )
Netherlands	Surface water → Coagulation → Sedimentation → disinfection (UV, O <sub>3</sub> , UV/H <sub>2</sub> O <sub>2</sub> ) → Activated carbon filtration → Sand filtration	<a href="#">van Dooremalen et al. (2020)</a>
Germany	Ground water → Aeration → Sand filtration	<a href="#">Zhang (2010)</a>
France	Groundwater/ Surface water → Coagulation → Sedimentation → Sand filtration → O <sub>3</sub> /BAC → Disinfection	<a href="#">Wang (2006)</a>
Israel and Saudi Arabia	Seawater → Pre-chlorination → Coagulation → Sedimentation → Sand filtration → Multistage flash/ Reverse osmosis → Disinfection	<a href="#">Al-Mutaz (1996);</a> <a href="#">Al-Jaseem et al. (2016);</a> <a href="#">Zhao et al. (2010)</a>
Singapore	(1) Reclaimed water → Micro filtration → Reverse osmosis → Ultraviolet (2) Seawater → Coagulation → Sedimentation → Sand filtration/Ultra filtration → Reverse osmosis	<a href="#">Bai et al. (2020), Tong et al., 2007</a>
China	Surface water → Coagulation → Sedimentation → Sand filtration → Chlorine disinfection	<a href="#">China Urban Water Association, 2019</a>
Japan	Surface water → Rapid sand filtration → O <sub>3</sub> /BAC → Chlorine disinfection	<a href="#">(Japan Water Works Association, 2017)</a>
USA	Surface water → Coagulation → Sedimentation → Sand filtration → O <sub>3</sub> /BAC → Chlorine disinfection	<a href="#">Yao et al. (2013)</a>
Australia	(1) Surface water → Coagulation → Sedimentation → Sand filtration → Chlorine disinfection (2) Seawater → Pre-chlorination → Coagulation → Sedimentation → Sand filtration → Reverse osmosis → Chlorine disinfection	<a href="#">(El Saliby et al., 2009); Chen (2003)</a>
South Africa	Surface water → Coagulation → Sedimentation → Sand filtration → Chlorine disinfection	<a href="#">Wang et al. (2017)</a>

removal of pollutants by oxidizing ozone molecules and free radicals (such as hydroxyl radicals, superoxide radicals, etc.) generated by ozone chain decomposition. The ozonation process can be classified into pre-ozonation and post-ozonation phases. The former is usually applied in the pretreatment of raw water while the latter is applied in advanced treatments.

Pre-ozonation of two waterworks located in Suzhou City, China, exhibited a poor reduction of total ARGs ([Guo et al., 2014](#)). Most of the ARGs were recalcitrant with small fluctuations. The absolute abundance of *tetX* gene was the only one that was reduced by approximately 1.8 log in one of the two waterworks. Another study documented that the pre-ozonation process slightly reduced the relative abundance of ARGs including aminoglycoside,  $\beta$ -lactam, sulfonamide, FCA, MLSB, other/efflux, vancomycin and tetracycline resistance genes but, increased absolute abundance of these ARGs ([Xu et al., 2016](#)). The poor reduction in ARGs may be attributed to the competitive consumption of the coexisting matrix constituents such as natural organic matter.

## 2.2. Conventional drinking water treatment processes

The conventional drinking water treatment processes in waterworks involve coagulation/sedimentation, sand filtration and chlorine/UV disinfection. Coagulation and sedimentation are two independent processes, while clarification integrates the two processes. During flocculation, impurities such as colloids, suspended solids, microorganisms, heavy metals, and organic pollutants in the raw water are aggregated into flocs that then settle down by gravity. The supernatant then flows into the sand filter and the residual tiny suspended mat-

ter, organic matter, and bacteria are further removed. Finally, the treated water is disinfected to further remove residual pathogenic microorganisms.

### 2.2.1. Coagulation/sedimentation and clarification

[Zhang et al. \(2016b\)](#) found that the coagulation and sedimentation processes of a waterworks in the Yangtze River Delta suppressed the absolute abundance of ARGs about 0.9 log. Other studies on waterworks located in the Yangtze River Delta and Qiantang Delta documented that coagulation/sedimentation and clarification units decreased the absolute abundance of most ARGs to a limited extent ([Guo et al., 2014; Xu et al., 2016](#)). [Wang et al. \(2019\)](#) reported that the absolute abundance of MCR-1 and NDM-1 increased by 0.42 log and 0.28 log respectively after flocculation. The sedimentation unite of two waterworks in Guangzhou, China, showed that, in one of them, the absolute abundance of most ARGs was increased while the absolute abundance was reduced in the other ([Su et al., 2018](#)). [Hu et al. \(2019\)](#) in their study documented that the total absolute abundance of ARGs was effectively reduced (1.6 log) by coagulation and sedimentation unites of a waterworks in East China.

### 2.2.2. Sand filtration

Several studies have been performed to ascertain the efficiency of ARGs removal by sand filtration. In one of them, it was found that sand filtration reduced the absolute abundance of ARGs and effectively decreased the relative abundance of ARGs ([Zhang et al., 2016b](#)). Due to the difficulties of cutting down the relative abundance, it was suggested that sand filtration be listed as a control technology

for ARGs. Another two works also inclined to this view that sand filtration could remove ARGs effectively (Su et al., 2018; Zheng et al., 2018). However, different outcomes were obtained from other studies. Two studies announced that the absolute abundance of ARGs decreased while the relative abundance increased after sand filtration (Guo et al., 2014; Xu et al., 2016). Another study declared that sand filtration was powerless to significantly reduce the relative and absolute abundance of MCR-1 and NDM-1 (Wang et al., 2019). In addition, the total absolute abundance of ARGs was shown not to have changed after flowing through the sand filter following flocculation/coagulation+sedimentation or pre-zonation +flocculation+sedimentation processes (Hu et al., 2019).

#### 2.2.3. Chlorine disinfection

Chlorination is an effective process for reducing the concentration of ARGs. The removal rate of ARGs has been positively correlated with the CT value of chlorine (chlorine dosage  $\times$  contact reaction time) (Destiani and Templeton, 2019; Stange et al., 2019a). Besides remarkable abatement of absolute abundance, raising of chlorine dose yielded a remarkable reduction of ARGs types (Lin et al., 2016a). Compared to ARB, ARGs are more difficult to remove. At the chlorine dose of 0.25 mg/L, the inactivation of ARB reached 5.1 log. However, only 0.4–0.5 log ARGs removal was attained (Stange et al., 2019a). It has also been shown that chlorination is effective and superior compared to the other water treatment processes (Guo et al., 2014; Xu et al., 2016). Nevertheless, the selective enrichment of some ARGs due to chlorination needs further studies (Guo et al., 2014; Shi et al., 2013; Xu et al., 2016; Zheng et al., 2018).

#### 2.2.4. Ultraviolet disinfection

Ultraviolet disinfection has been shown to be inferior to chlorination with regards to ARGs removal. Utilizing multiple antibiotic resistant *E. coli* and *P. aeruginosa* as test bacteria, a UV<sub>254</sub> fluence of 200 mJ/cm<sup>2</sup> resulted in 1.2 log ARGs reduction while chlorination at a lower dose (30 mg/L•min) achieved 1.7 log reduction (Destiani and Templeton, 2019). An increase in the UV dose over the suggested values (40–200 mJ/cm<sup>2</sup>) did not enhance the reduction efficiency. For instance, a UV<sub>254</sub> dose of 600 mJ/cm<sup>2</sup> inhibited ARGs hosting in *E. faecium* by 0.33 log (Stange et al., 2019a) while the dose of 400 mJ/cm<sup>2</sup> slightly enhanced the elimination of ARGs hosting in SER2 strain and MAR SER6-1 strain compared to the UV<sub>254</sub> dose of 200 mJ/cm<sup>2</sup> (Zhang et al., 2017a). Because of its lack of continuous disinfection abilities, UV is normally used in tandem with chlorination (Zhu and Chen, 2013). The combination of UV disinfection and chlorination exhibited a better ARGs removal performance compared to chlorination or UV disinfection alone (Destiani and Templeton, 2019; Zhang et al., 2019c, 2015).

### 2.3. Advanced drinking water treatment processes

Advanced treatment processes (ATPs) are performed after the conventional treatment processes to further improve water quality. ATPs decrease the levels of the remaining pollutants including natural organic matter, micro-polluted organic matter, and pathogenic microorganisms. Ozonation, in combination with biological activated carbon (O<sub>3</sub>/BAC) and membrane

filtration are the two widely used ATPs in waterworks. The membrane filtration process includes microfiltration (MF), ultrafiltration (UF), nanofiltration (NF), and reverse osmosis (RO). The most common membrane filtration processes in waterworks are UF and RO (Slipko et al., 2019). UF effectively removes macromolecular organics, suspended solids, colloids, bacteria, and viruses from water, thereby providing the possibility of removing ARB and ARGs (Fan et al., 2013; Jiang et al., 2003; Krzeminski et al., 2020).

#### 2.3.1. Post-ozonation

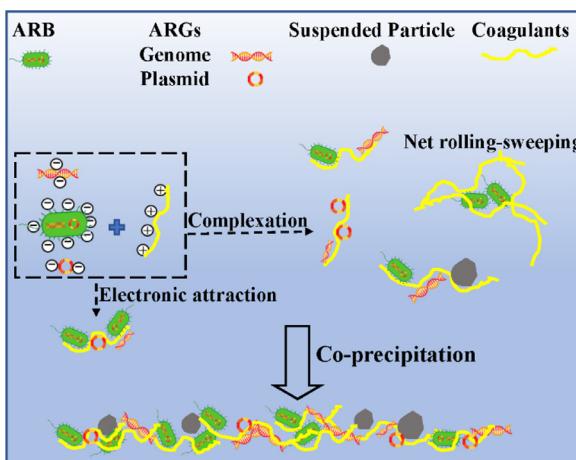
Reduction in the concentration of ARGs by post-ozonation, has not been conclusively studied. Studies have documented an increased absolute abundance of several ARGs (tetA, tetM, tetW, cfr, cmlA, qnrB, qnrS) after post-ozonation (Su et al., 2018). However, a decreased absolute abundance of target ARGs after post-ozonation has also been documented (Xu et al., 2016; Zheng et al., 2018). Guo et al. (2018) reported that in waterworks JS and JG, post-ozonation exhibited a significantly positive contribution towards the reduction in the absolute concentrations of ARGs and a negative contribution towards the reduction in the relative abundance of tetracycline resistance genes. Furthermore, post-ozonation in waterworks SX exhibited a little effect on the relative and absolute abundance of ARGs. Post-ozonation in waterworks W significantly reduced both relative and absolute abundance of ARGs.

#### 2.3.2. Biological activated carbon process

The BAC process utilizes the adsorptive ability of activated carbon and the biodegradation ability of biofilm on the activated carbon surface to remove pollutants. The ARGs removal efficiency by the BAC process differs among different waterworks. In four waterworks (waterworks W, waterworks SX, waterworks JS, and waterworks JG) in the Yangtze River Delta (Guo et al., 2014), it was found that 80%–90% target ARGs in waterworks W and SX were enriched by BAC while the ARGs could not be detected by qPCR in waterworks JS after the BAC process. In waterworks JG, tetO, tetM, and tetW genes were the only ones whose absolute abundance decreased by less than 1 log, tetC gene increased by about 1.27 log while the other ARGs were slightly decreased. This enrichment effect (increase of relative abundance) by the BAC process has also been documented by other studies (Xu et al., 2016; Zheng et al., 2018). The BAC process can also change the type of ARGs. A previous study showed that after treatment by the BAC process, the ARGs type increased from 84 to 159 (Zheng et al., 2018).

#### 2.3.3. Membrane filtration process

The membrane filtration process including UF, NF, and RO processes exhibited a good performance in ARGs elimination when a proper membrane molecular weight cut off (MWCO) was selected. Guo et al. (2014) showed that UF membranes with MWCO 20–100 KDa in waterworks was unable to effectively eliminate ARGs. Instead, the abundance of most ARGs in the effluent increased by varying degrees. In bench scale tests, UF, NF, and RO processes performed well in the elimination of plasmid-encoded ARGs dispersed in ultrapure water. The plasmid removal rate increased with a decrease in membrane MWCO (Krzeminski et al., 2020). The UF membrane with 1 KDa MWCO can intercept more than 2 log plasmids. For the



**Fig. 1 – Mechanism of ARGs reduction during coagulation/sedimentation or clarification treatment.**

UF membrane with MWCO 300 KDa, only 23% of the plasmid ( $0.07\text{ }\mu\text{m}$ ) was retained (Slipko et al., 2019). When the MWCO of the membrane (UF, NF, and RO) was less than 5 KDa, more than 99.80% aqueous plasmids and linear DNA carrying ARGs were removed.

### 3. Removal mechanisms of ARGs by different drinking water treatment processes

#### 3.1. Coagulation/sedimentation and clarification processes

Most ARGs are carried by ARB. Coagulation/sedimentation and clarification processes achieve a reduction of intracellular ARGs by removing ARBs through electrostatic attraction (negatively charged viruses + positively charged iron/aluminum oxyhydroxide floc particles) (Zhang et al., 2016b). Extracellular ARGs can be directly bound to hydrolyzed coagulants and settled with flocs (Fig. 1). Thus, background constituents, coagulant types, extracellular ARGs fragment size, and ARB type can influence ARGs removal (Zhuang et al., 2014). Notably, if the vigorous hydraulic agitation caused the sedimentation sludge containing ARGs resuspend, the abundance of ARGs in the effluent may increase (Hu et al., 2019).

#### 3.2. Sand filtration

Sand filtration removes ARGs through the contact adhesion effect (Fig. 2). This is attributed to the small size of ARB and free ARGs when compared to the pore size of the filter bed (Su et al., 2018). In long-term operation of the sand filter, biofilms grow on the surface of filter materials. Biofilms facilitate HGT of ARGs. However, in a nutrient-poor environment, ARB are prone to losing resistant plasmids and may therefore grow with a gradual loss of antibiotic resistance (Griffiths et al., 1990). Aging biofilms fall off from the surface of the filter media, enter into the bulk water and increase the abundance of ARGs (Wang et al., 2017b).

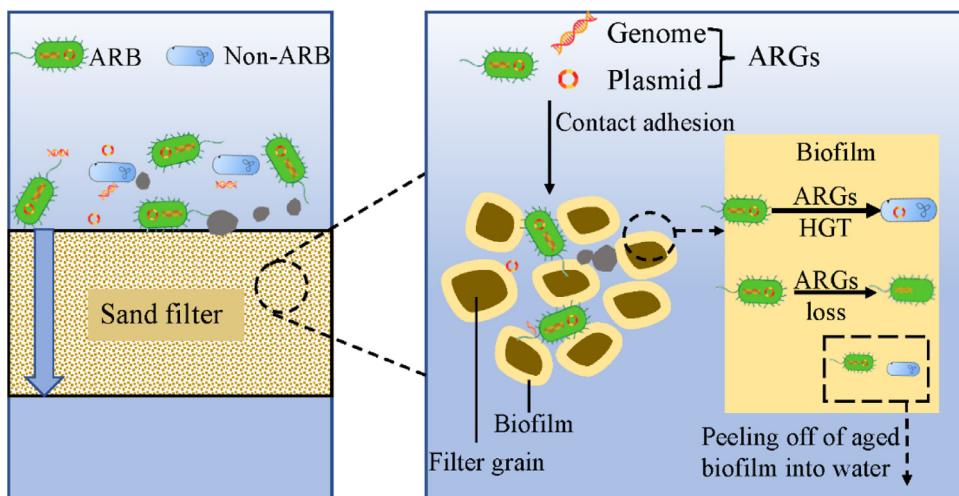
### 3.3. Ozonation process

The mechanism of ARGs reduction by ozonation process is presented in Fig. 3. Ozone is a strong oxidant with an oxidation potential of 2.07 V. It can decompose through chain reactions to generate hydroxyl radicals ( $\text{HO}\cdot$ ) that are non-selective and strongly oxidative. Ozone and  $\text{HO}\cdot$  can damage extracellular DNA molecules in water (Fan, 2017; Qu et al., 2013). However, concerning intracellular DNA, water background components, cell wall/membrane molecules, and cytoplasmic constituents may take precedence over DNA molecules to consume ozone and  $\text{HO}\cdot$  (Zhang et al., 2019c). Ozone is strongly reactive in the presence of amino acids and unsaturated carbon-carbon bonds in proteins, peptidoglycans and lipids. Therefore, it causes functional failures of cell wall/membrane (such as cell permeability increase) and ARGs-carrying DNA, or destroys their structure (Dodd, 2012; Ren et al., 2011). The destruction of the cell wall/membrane causes transformations of intracellular ARGs to extracellular ARGs. The latter exhibits a better chance coming into contact with ozone and  $\text{HO}\cdot$ , leading to a more efficient removal of ARGs.

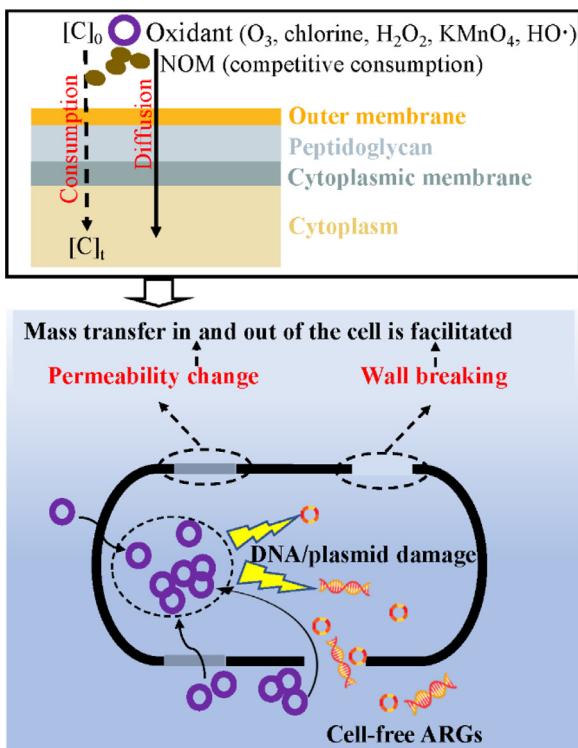
The reactions between intracellular ARGs with ozone and its secondary oxidant  $\text{HO}\cdot$  cannot be initiated until the oxidants penetrate into the cell and achieve a certain cumulative concentration. Higher ozone doses may, therefore, be needed. Studies have shown that increasing the exposure time and ozone dose to a certain range enhances the elimination of ARGs (Iakovides et al., 2019; Oh et al., 2014; Sousa et al., 2017; Stange et al., 2019a). This mechanism can be used to explain why the pre-ozonation process ( $0.3\text{--}0.6\text{ mg/L O}_3$ ) is less effective than the post-ozonation process ( $1.5\text{--}1.7\text{ mg/L O}_3$ ) (Xu et al., 2016). However, some studies found that increasing the ozone dose does not enhance the efficiency of ARGs removal (Zheng et al., 2017). In addition, some studies found that the ozonation process increased the absolute abundance of ARGs in the effluent (Guo et al., 2014; Su et al., 2018). Su et al. (2018) speculated that the regulatory locus *soxRS* induced the increase in antibiotic resistance of ARB under the selective pressure of ozone.

#### 3.4. Biological activated carbon process

Biological activated carbon adsorbs and enriches antibiotics to induce the generation of ARGs (Xu et al., 2016). Activated carbon can also intercept ARGs-hosting bacteria and reduce the concentration of ARGs in the water phase. Biofilms are composed of proteins, polysaccharides, DNA, RNA, peptidoglycan, lipids, and phospholipids (Høiby et al., 2010). These compounds enhance the growth and remediation of bacteria on the biofilm. Biofilm formation is affected by quorum sensing. Quorum sensing refers to bacterial community behaviors that rely on self-induced factors (usually called signal molecules) to sense the amount or density of bacteria around them and regulate gene expression (Liu et al., 2012). For example, acyl homoserine lactones (AHLs) is a signal molecule with little effect on bacterial growth but can promote the horizontal transfer of resistant plasmids, thereby intensifying the spread of antibiotic resistance (Zheng et al., 2018). Horizontal gene transfer may also occur between the water phase and the biofilm phase (Parkas et al., 2013). Peeling off of aged biofilms



**Fig. 2 – Mechanism of ARGs reduction and dissemination during sand filtration treatment.**

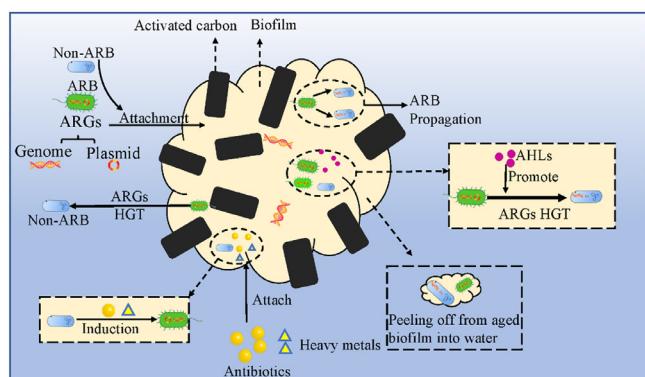


**Fig. 3 – Mechanism of ARGs reduction during ozonation treatment.**

into water exhibits a negative effect on the removal of ARGs removal by BAC (Hu et al., 2019). Due to the difficulties brought by the differences in feeding water quality and biofilm property in different waterworks, the current research on mechanisms of ARGs removal by BAC (Fig. 4) remains at a relatively shallow level and needs in-depth study.

### 3.5. Membrane filtration process

The membrane filtration process eliminates water pollutants through size exclusion, electrostatic effect, and diffusion ef-



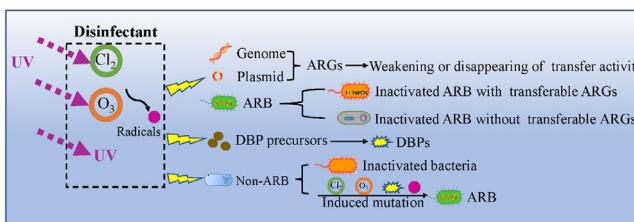
**Fig. 4 – Mechanism of ARGs reduction and dissemination during BAC treatment.**

fect (Ernst et al., 2000; Ganiyu et al., 2015; Taheran et al., 2016). Ultrafiltration effectively eliminates ARB and is believed to be efficient in the removal of intracellular ARGs (Lan et al., 2019). Most of the extracellular ARGs exist on mobile genetic elements (MGEs), such as transposons, integrases, and plasmids (Guo et al., 2014). Plasmids exhibit certain flexibilities in size and shape (Slipko et al., 2019). A plasmid can be stretched and elongated in the converging flow field and dovetail through pores that are smaller than its radius (Morão et al., 2011; Slipko et al., 2019). Linear DNA molecules exhibit better flexibility when compared to plasmids. This property enhances their penetration through membranes (Latulippe and Zydny, 2011). A decrease in the MWCO of the membrane reduces the ability of plasmids to pass through the membrane and enhances its interceptive success rate for ARGs. Slipko et al. (2019) documented that membranes with MWCO  $\leq 2500$  Da should be applied in the treatment of drinking water to enhance DNA interception.

In addition to the sieving effect that is associated with the membrane pore size, charge interactions between the surface of the membrane and ARGs are involved in ARGs removal mechanisms (Fig. 5). Deoxyribonucleic acid molecules are



**Fig. 5 – Mechanism of ARGs reduction during membrane filtration treatment.**



**Fig. 6 – Mechanism of ARGs reduction during disinfection treatment.**

negatively charged. Therefore, when the surface of the membrane is negatively charged, electrostatic repulsion enhances the retention of free DNA (Ager et al., 2009). However, an increase in driving forces such as transmembrane pressures may lead to the ARGs rejection to fall, not rise. This effect could be attributed to overcompensation of the electrostatic repulsion by the convergent flow field (Slipko et al., 2019). However, the above mentioned ARGs removal mechanism does not provide a plausible explanation for the increased ARGs abundance in the effluents from the UF process as found in some studies (Guo et al., 2014).

### 3.6. Disinfection process

The disinfection processes inactivate ARB by destroying their cellular structure or DNA (Fig. 6). However, ARGs have been known to survive in cellular debris. Chlorination is widely used because of its efficient and continuous disinfection ability (Deborde and von Gunten, 2008). Electrically neutral hypochlorous acid molecules are easily close to the negatively charged surface of the cell and diffuse into the cell. Then, the penetrated active chlorine damages bacterial enzymes and leads to a sugar metabolism disorder that inactivates bacteria. However, before the active chlorine molecules interact with enzymes and ARGs-encoding DNA, it is consumed by components of the cell wall/membrane and the cytoplasm (Dodd, 2012; Sharma et al., 2016). For efficiency in the removal of ARGs, it is important to ensure sufficient exposure to chlorine (chlorine dose × contact time) (Chen and Zhou, 2018; Dodd, 2012). This aspect is attributed to the enrichment of some types of ARGs borne in chlorine-tolerant bacteria (Huang et al., 2011; Jia et al., 2015; Shi et al., 2013; Xi et al.,

2009). Co-resistance of disinfectants and antibiotics mediated by multidrug resistance efflux pumps has been associated with chlorine resistance (Yuan et al., 2015). In addition, the increased abundance of MGEs and excessive disinfectant-stimulated plasmid replication may also be attributed to the enrichment of ARGs (Shi et al., 2013). Bacteria acquire co-resistance to disinfectants and antibiotics and proliferate in clean water reservoirs to enhance the abundance of ARGs (Xu et al., 2016).

Ultraviolet light penetrates the relatively UV-transparent bacterial cell envelope and cytoplasm, where it is absorbed by the pyrimidine and purine bases that assemble into DNA and RNA. The DNA molecules of pathogenic microorganisms absorb UV light in the ranges of 200~280 with a maximum absorption peak at 254 nm (Nebot Sanz et al., 2007). Ultraviolet damages the molecular structure of DNA by inducing the formation of pyrimidine dimers or by inducing DNA strand breakage. These effects inhibit the ability of the bacteria to reproduce and finally leads to cell death (Dodd, 2012; Nebot Sanz et al., 2007; Zhang et al., 2004). Ultraviolet irradiation can, therefore, induce direct photolytic degradation and deactivation of intracellular ARGs-encoding DNA. However, even if DNA molecules are damaged, ARGs fragments may remain intact (Stange et al., 2019a). In addition, it has been reported that a large number of spores contain traceable amounts of ARGs formed after UV disinfection (Zheng et al., 2017). Under favorable conditions the spores can develop into mature microorganisms (Michod et al., 2008).

Ultraviolet /chlorination (UV/Cl<sub>2</sub>) disinfection is characterized by the formation of non-selective HO• and selective reactive chlorine species (RCSs, including Cl•, Cl<sub>2</sub>•<sup>-</sup> and ClO<sup>•</sup>) (Fang et al., 2014; Feng et al., 2007). Before interacting with ARGs, HO• and RCSs have to pass through cell membrane and cytoplasmic components. Zhang et al. (2019c) found that RCS was more efficient than HO• due to the fast consumption of HO•. Three possible mechanisms are involved in ARGs elimination by the UV/chlorination disinfection process: i. UV irradiation damages the cell wall and cell membranes, enhancing the entry of chlorine and free radicals into the cell that then react with ARGs (Zhang et al., 2006); ii. Chlorine and free radicals damage the cell membrane and facilitates the adsorption of UV into the cells to destroy ARGs-encoding DNA molecules (Destiani and Templeton, 2019; Li et al., 2018a) and iii. cell permeability is damaged under the actions of UV/chlorine, followed by the discharge of intracellular ARGs-encoding DNA and the formation of cell-free ARGs that are degraded by UV, chlorine, and other radicals (Dodd, 2012).

## 4. Factors affecting the removal of ARGs during water treatment processes

### 4.1. Abiotic factors

#### 4.1.1. Selective pressures

Selective pressures from antibiotics, heavy metals (Baker-Austin et al., 2006; Zhang et al., 2018a), disinfectants (Kim et al., 2018; Zhang et al., 2017b), and disinfection by-products (Li and Gu, 2019) in the water treatment process increase the risk of ARGs and ARB enrichment as well as ARGs

removal. The emergence of ARGs is mainly attributed to antibiotics. Antibiotic concentration in drinking water is relatively low (usually at ng/L). These low concentrations are, therefore, difficult to induce the generation of ARGs and the selection of ARB (Lv et al., 2014). However, as mentioned earlier, the BAC process can adsorb and enrich antibiotics thereby creating a suitable environment for the development of antibiotic resistance. Heavy metals also stimulate the emergence of bacterial resistance. Ji et al. (2012) revealed that the concentration of heavy metals (copper, mercury, zinc) in the soil had a strong positive correlation with the concentration of ARGs. Yang et al. (2017) documented a similar finding after their studies in the lakes of Wuhan. Stepanauskas et al. (2006) documented that heavy metals could select for multi-drug resistant bacteria in fresh water while Zhang et al. (2018b) showed that four heavy metals including Cu(II), Ag(I), Cr(VI), and Zn(II) promote HGT of ARGs under sub-inhibitory concentrations. Due to the low concentrations of heavy metals in drinking water, it has not been established if concentrations below the recommended levels can screen for ARB.

Unlike antibiotics and heavy metals, the concentrations of disinfectants in drinking water is relatively high (usually mg/L). Chlorine disinfectants have been shown to change the bacterial community structure and enrich ARGs and ARB in drinking water plants. Chlorine reacts with humic acid, algae, and other organics to generate disinfection by-products (DBPs) while at the same time killing bacteria. Disinfection by-products enhance bacterial resistance. Four typical DBPs, 3-chloro-4-(dichloromethyl)-5-hydroxy-2(5H)-furanone (MX), dichloroacetonitrile, potassium bromate, and dibromoacetic acid, were shown to enhance *Pseudomonas aeruginosa* PAO1 resistance to 10 antibiotics by various levels (Lv et al., 2014). These compounds enhanced antibiotic resistance by a factor of more than 10 when compared to the control. The induction of antibiotic resistance by MX was the most conspicuous and was shown to be realized through mutagenesis that enhanced the overexpression of efflux pumps. Through a similar mechanism, DBPs such as trichloroacetic acid, chlorite, iodoacetic acid, bromoacetamide, trichloroacetonitrile, and tribromonitromethane were also found to enhance bacterial resistance (Li et al., 2016; Lv et al., 2015).

Under multiple environmental selective pressures, bacteria develop co-resistance or cross-resistance (Zhang et al., 2016a). Antibiotic resistance bacteria with co-resistance to chlorine and antibiotics have been found to be the dominant strains after disinfection associated pressures (Jia et al., 2015; Shi et al., 2013). Yang et al. (2017) documented that the co-selection mechanisms of heavy metals and antibiotics promote the emergence and spread of ARGs in lake water. Therefore, more studies should be done to elucidate on co-resistance and cross-resistance.

#### 4.1.2. Nutrition limiting factors

Selective pressure is the basis for maintaining bacterial resistance. When the selective pressure is high, the bacteria with plasmid encoded resistance mechanisms to the source of the pressure have a growth advantage. However, ARB require additional energy to express resistance and, therefore, lose a competitive advantage with wild-type sensitive flora. This phenomenon is referred fitness cost. Lin et al. (2016b) found that

the lower the concentration of total organic carbon (TOC), the more evident the fitness cost and the faster the loss of resistant plasmids. However, the degree of fitness cost varies among different types of plasmids. This implies that fitness cost is relevant to nutritional levels besides selective pressures. Griffiths et al. (1990) reported that resistant plasmids can be partially or completely lost in a poor nutritional environment. This effect leads to a loss in the ability of bacteria to resist antibiotics. A study performed by Lin et al. (2018) found that antibiotic resistance increases fitness cost only under high nutrient conditions. Poor nutrient conditions weaken the fitness cost. This is attributed to the adaptive mutations and low relative expression of the *rpoS* gene. In addition, Wan et al. (2019) also found that decreasing the TOC concentration also increases the relative abundance of ARGs (that is, decreasing the fitness cost).

Organic carbon, nitrogen and phosphorus are nutritional factors required for bacterial growth. Ali et al. (2016) revealed that the addition of nitrogen and phosphorus to bacterial growth medium enhanced the resistance of *E. faecalis* to oxytetracycline (OXY), chloramphenicol (CHL), vancomycin and erythromycin (ERY). However, only a high dose of the nutrients led to ciprofloxacin (CIP) resistance. Studies have also observed that the abundance of ARGs in freshwater lakes or rivers is positively correlated with the concentration of nitrogen and phosphorus (Wang et al., 2020; Xi, 2018). Eutrophic water is beneficial for the propagation of ARGs. Therefore, regulating the concentrations of nitrogen and phosphorus in drinking water is important in the inhibition of the spread of drug resistance.

## 4.2. Biotic factors

### 4.2.1. Biofilms

Biofilms are composed of microorganisms and extracellular polymers (EPS). Extracellular polymers refer to polymers, proteins, and ribose substances. Extracellular polymers enhance the clustering property of biofilms and their firm adsorption on the surface of the carrier. More importantly, EPS provide a "shelter" for the microorganisms in the biofilm to resist external harm such as water shear force, antibiotic activities and oxidant effects (Fish et al., 2017; Flemming, 2009). Biofilms also exhibit advantages in biodiversity, gene pool richness, genetic exchange convenience, biocides and other stress resistance abilities, nutrients recycle, and high-density population (Flemming, 2009). These ecological advantages explain the relatively frequent ARGs transmission and ARB propagation in the biofilm.

Biofilms are common in DWTPs, especially BAC filter. Due to the ideal environment they provide and their stable structure, biofilms serve as reservoirs for the spread of ARGs and ARB. They lead to poor ARGs removal efficiency of BAC (Bai et al., 2015).

### 4.2.2. Types of ARB and ARGs

The removal efficiency of ARGs is associated with the types of ARB. When ARB resist external interference or destruction, it is difficult to eliminate the ARGs in the ARB. Jia et al. (2015) found that ARGs in chlorine-tolerant bacteria

such as *Methylophilus* could be effectively removed by chlorination, while ARGs in strong chlorine-tolerant bacteria such as *Pseudomonas* and *Acidovorax* could survive and become the dominant species. [McKinney and Pruden. \(2012\)](#) documented that gram-positive ARB exhibited better tolerance to ultraviolet disinfection compared to gram-negative ARB. The small total genome sizes of the gram-positive ARB reduced the number of potential pyrimidine dimer targets, thereby leading to the low susceptibility to UV.

The nature and structure of different types of ARGs can also affect the efficiency of their removal. [Xu et al. \(2016\)](#) found that among the MLSB resistance genes in BAC tank effluents, the *erm* genes were easily captured by MGEs and could be readily transferred among different host bacteria compared to the other genes. This property of *erm* genes enhances their persistence in drinking water treatment flow. [Destiani and Templeton. \(2019\)](#) revealed that the removal efficiency of different ARGs by UV disinfection was consistent with the number of potential dimers, particularly TT dimers on each gene.

#### 4.2.3. Existence form of ARGs

Antibiotic resistance genes are commonly divided into cell-carrying ARGs (intracellular ARGs) and cell-free ARGs (extra-cellular ARGs). Without the protection of cell walls and cell membranes, extracellular ARGs are easily damaged by UV irradiation and oxidants such as chlorine ([Yoon et al., 2017](#)). However, cell-free ARGs easily penetrate UF and RO membranes due to their small sizes compared to cell-carrying ARGs.

Environmentally, ARGs can be classified as particle-associated ARGs, biofilm-associated ARGs, and free ARGs in water ([Zhang et al., 2019a](#)). Particle-associated ARGs are easily ignored. Particles serve as “seeds” for the attachment of ARGs and ARB that lead to the formation of biofilms. Under the shield of particles and biofilms, the removal of ARGs by chlorination, ozonation, and ultraviolet irradiation is inhibited ([Liu et al., 2013](#)). However, sand and membrane filtration processes may effectively intercept particle-associated ARGs.

## 5. Summary and outlook

In conclusion, it is commonly accepted that drinking water treatment process, excellent in removing or separating host bacteria of ARGs from water, is more promising on ARGs removal. The comparisons of the removal efficiency of ARGs by different water treatment processes found that: i. coagulation, sedimentation/clarification, sand filtration, and BAC filtration cannot effectively remove ARGs; ii. The effect of ozonation process is uneven; iii. The ARGs removal efficiency by membrane filtration depends on the MWCO and iv. The disinfection process is effective; however, chlorination may enrich ARGs while UV disinfection requires a high UV dosage. In contrast, a combination of UV/chlorination is more reliable.

The ARGs removal efficiency by drinking water treatment processes is influenced by diverse factors such as selective pressure, nutrition level, ARGs types and existing form, as well as processing parameters. The combined effects of various factors complicate the ARGs removal process. Generally, knowledge regarding the removal mechanisms of ARGs by

different drinking water treatment processes is at the initial stages. Future studies should focus on:

- (i) The removal properties and mechanisms of ARGs of different types or forms in specific drinking water treatment processes.
- (ii) The effects and mechanisms of the disinfection processes and disinfection by-products on the generation and spread of various ARGs.
- (iii) The post-treatment of ARGs in sludge and the concentrate from membrane filtration generated in waterworks to avoid secondary pollution.
- (iv) Exploration of rapid analysis methods to quantify the transmissibility of ARGs in drinking water, rather than simple concentration evaluation.

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